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Growth of nucleation mode particles in the summertime Arctic: a case study

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Abstract. The summertime Arctic lower troposphere is a relatively pristine background aerosol environment dominated by nucleation and Aitken mode particles. Understanding the mechanisms that control the formation and growth of aerosol is crucial for our ability to predict cloud properties and therefore radiative balance and climate. We present an analysis of an aerosol growth event observed in the Canadian Arctic Archipelago during summer as part of the NETCARE project. Under stable and clean atmospheric conditions, with low inversion heights, carbon monoxide less than 80 ppbv, and black carbon less than 5 ngm⁻³, we observe growth of small particles, < 20 nm in diameter, into sizes above 50 nm. Aerosol growth was correlated with the presence of organic species, trimethylamine, and methanesulfonic acid (MSA) in particles ~ 80 nm and larger, where the organics are similar to those previously observed in marine settings. MSA-to-sulfate ratios as high as 0.15 were observed during aerosol growth, suggesting an important marine influence. The organic-rich aerosol contributes significantly to particles active as cloud condensation nuclei (CCN, supersaturation = 0.6 %), which are elevated in concentration during aerosol growth above background levels of ~ 100 to ~ 220 cm⁻³. Results from this case study highlight the potential importance of secondary organic aerosol formation

and its role in growing nucleation mode aerosol into CCN-active sizes in this remote marine environment.

1 Introduction

In the warming Arctic (Jeffries and Richter-Menge, 2012), decreasing sea ice extent (Lindsay et al., 2009) is likely to increase the oceanic influence on atmospheric composition. This change in exposed ocean area will have implications on aerosol concentrations and composition and therefore on cloud properties (Browse et al., 2014) and precipitation (Kopeck et al., 2016). Aerosol–cloud–climate interactions are unique in Arctic regions due to the high surface albedo, the seasonal cycle in aerosol loading and properties, the strong static stability in the lower troposphere (Aliabadi et al., 2016b), and the dependence of cloud infrared emissivity on droplet size and aerosol characteristics (Curry, 1995).

Pristine background aerosol conditions prevail in the summertime Arctic boundary layer. A pronounced seasonal cycle characterizes Arctic aerosol (Engvall et al., 2008; Sharma et al., 2013; Tunved et al., 2013; Croft et al., 2016; Nguyen et al., 2016), with strong anthropogenic contributions to “Arctic haze” in winter and spring (Law and Stohl, 2007; Quinn et al., 2007) and more regional influences in the

cleaner summer months, especially in the lower troposphere (Leaitch et al., 2013; Heintzenberg et al., 2015). Beginning in late spring, efficient wet removal of aerosol and less efficient transport from lower latitudes come together to suppress the condensation sink (Stohl, 2006; Engvall et al., 2008) and allow nucleation and Aitken mode particles to dominate the size distribution (Engvall et al., 2008; Heintzenberg and Leck, 2012; Croft et al., 2016). Under these clean conditions, cloud condensation nuclei (CCN) and cloud droplet number concentrations can be exceptionally low (Mauritsen et al., 2011; Leaitch et al., 2016), making summertime liquid clouds very sensitive to the formation of new particles and their growth into CCN sizes. Since Arctic clouds are an important determinant of the local surface energy balance (e.g., Intrieri et al., 2002; Lubin and Vogelmann, 2006) and have the ability to influence the thickness, freezing, and melting of sea ice (Kay and Gettelman, 2009; Tjernström et al., 2015), a predictive understanding of the sources and processes controlling CCN-active aerosol is a crucial aspect of understanding the Arctic climate.

While transport of pollutants from lower latitudes does occur in Arctic summer, especially in the middle and upper troposphere, efficient scavenging during transport and within Arctic regions results in an important contribution from regional sources near the surface at this time of year (e.g., Stohl, 2006; Garrett et al., 2011; Croft et al., 2016). In the absence of significant transported aerosol, several different processes can contribute to aerosol formation, including the emission of primary particles from the ocean surface, along with formation of new particles by nucleation and their subsequent growth by condensation and coagulation.

The formation of new particles can be an important aerosol source in the summertime Arctic (Leaitch et al., 2013; Croft et al., 2016). Through its oxidation to sulfuric acid and other products, dimethyl sulfide (DMS) plays an important role in the formation, and growth, of new particles (Leaitch et al., 2013). In the Arctic and at midlatitudes, uncertainties in the rates and mechanisms of nucleation and growth are such that some studies are able to explain ambient observations with standard parametrizations developed from measurements at more southerly locations (e.g., Chang et al., 2011b), while others must invoke alternative mechanisms (e.g., Karl et al., 2012). The role of ammonia and amines in particle nucleation at midlatitudes has become well established (Almeida et al., 2013), and recent measurements suggest that local ammonia sources in the summer Arctic are sufficient to promote particle formation (Wentworth et al., 2016; Giamarelou et al., 2016). Iodine oxides can make a significant contribution to new particle formation in marine and coastal environments at midlatitudes (e.g., O'Dowd and de Leeuw, 2007); these species may contribute to the formation and growth of small particles in Arctic regions, although their biotic and abiotic sources in ice-covered regions remain unclear (Mahajan et al., 2010; Allan et al., 2015). Organic condensible species also play a role in nucleation, and growth, of particles

at midlatitudes (e.g., Kulmala and Kerminen, 2008; Metzger et al., 2010; Ehn et al., 2014; Tröstl et al., 2016); however, no direct evidence for the role of organic species in Arctic nucleation events exists to date.

The ejection of primary aerosol from the sea surface, through wave breaking and bubble bursting, is another source of aerosol across the size distribution (Ovadnevaite et al., 2014; Clarke et al., 2006; Nilsson et al., 2001). At midlatitudes a large organic fraction, which originates from the enrichment of biologically derived organic material at the sea surface, is frequently observed in marine aerosol (Facchini et al., 2008b; Russell et al., 2010; Gantt and Meskhidze, 2013; Frossard et al., 2014; Quinn et al., 2015a; O'Dowd et al., 2015; Quinn et al., 2015b). This primary marine organic aerosol (OA) tends to be water insoluble with chemical similarity to lipids (e.g., Rinaldi et al., 2010; Decesari et al., 2011) and has been demonstrated to have a source near the ocean surface (Ceburnis et al., 2008). Some similar observations have been made in Arctic regions (e.g., Narukawa et al., 2008; P. M. Shaw et al., 2010; Orellana et al., 2011; Fu et al., 2013; Karl et al., 2013; Fu et al., 2015). For example, Fu et al. (2013, 2015) have shown a dominance of primary saccharides and evidence for protein and humic-like substances in Arctic aerosol, suggesting an important local or regional source of primary marine OA. The release of marine micro-gels via bubble bursting in open leads has been proposed to contribute significantly to particles over the Arctic Ocean (e.g., Bigg and Leck, 2001; Orellana et al., 2011).

Particle growth through condensation of gas-phase species can also play a role in driving marine aerosol characteristics, making ambient marine OA a complex result of primary and secondary processes (e.g., Ceburnis et al., 2008; Facchini et al., 2008b; Rinaldi et al., 2010; Frossard et al., 2014). In contrast to primary marine OA, secondary marine OA is generally more water soluble and is composed of more-oxygenated compounds (Rinaldi et al., 2010; Decesari et al., 2011). Precursors of secondary marine OA include DMS and other biological volatile organic compounds (BVOCs), such as isoprene, monoterpenes, and amines, which are produced by a variety of marine micro-organisms (S. Shaw et al., 2010; Gantt et al., 2009; Facchini et al., 2008a). However, in the absence of specific molecular tracers it can be very challenging to discern the relative contribution of primary and secondary processes to ambient marine organic aerosol (e.g., O'Dowd et al., 2015). At midlatitudes, direct and indirect measurements of Aitken mode particle composition have demonstrated the role of secondary organic species in the growth of small particles (Vaattovaara et al., 2006; Bzdek et al., 2014; Lawler et al., 2014). Significant fractions of alkylamines, dicarboxylic acids, methanesulfonic acid (MSA), oxalic acid, alcohols, and other organic acids have been observed in marine aerosol, suggesting contributions from secondary processes (e.g., Facchini et al., 2008a; Claeys et al., 2010; Rinaldi et al., 2010; Dall'Osto et al., 2012; Frossard et al., 2014). In Arctic regions, the detection of specific molecu-

lar tracers for isoprene, terpene, and fatty acid oxidation has indicated a contribution of secondary processes to summertime organic aerosol (Fu et al., 2009; Kawamura et al., 2012; Fu et al., 2013; Hansen et al., 2014).

Our understanding of summertime Arctic aerosol remains incomplete, in part due to a scarcity of observations focusing on the influence of local and regional sources on aerosol chemical and physical properties. In this case study we focus on observations of a new particle formation and growth event made during the NETCARE summer aircraft campaign in July 2014, near Resolute Bay, Nunavut, Canada, in a general time period and location that was shown to have high biological activity in the surface ocean (Gosselin et al., 2015; Mungall et al., 2016). We use these observations to explore the composition and formation processes of particles contributing to CCN in the Canadian Arctic Archipelago during summer.

2 Methods

2.1 Measurement platform and inlets

As part of the NETCARE project (Network on Climate and Aerosols: Addressing Key Uncertainties in Remote Canadian Environments, <http://www.netcare-project.ca>), measurements of aerosol physical and chemical properties, trace gases, and meteorological parameters were made aboard the Alfred Wegener Institute (AWI) Polar 6 aircraft, a DC-3 aircraft converted to a Basler BT-67 (Herber et al., 2008). Measurements aboard Polar 6 took place from 4 to 21 July 2014, based in Resolute Bay, Nunavut (74°41'N, 94°52'W). The survey speed was maintained at approximately $\sim 75 \text{ m s}^{-1}$ for measurement flights, with ascent and descent rates of 150 m min^{-1} for vertical profiles.

The main aerosol inlet was located on the starboard side of the fuselage ahead of the engines. Based upon a total flow drawn to instruments of 35 L min^{-1} and a measured flow at the exhaust of the sampling line of 20 L min^{-1} , the total flow through the shrouded inlet diffuser was nearly isokinetic at 55 L min^{-1} . Aerosol flowed into the cabin through a stainless steel manifold (outer diameter of 2.5 cm, inner diameter of 2.3 cm) and was directed to the various particle instruments through stainless steel lines that branched from the main inlet at angles less than 90°. Aerosol was not dried prior to sampling; however, the temperature in the inlet line was approximately 10–15 °C warmer than the ambient temperature so that the relative humidity (RH) decreased significantly as the aerosol entered the sampling line. Exhaust from the main aerosol inlet flowed freely into the back of the cabin to keep the inlet from being over-pressured. Therefore, the total flow through the main aerosol inlet was dictated by the true airspeed (TAS). With the survey air speed noted above, transmission efficiency of aerosol through the main inlet was near unity for particles 20 nm to $\sim 1 \mu\text{m}$ in diameter.

Trace gases (CO , CO_2 , and H_2O) were sampled through a second inlet consisting of a 0.40 cm (outer diameter) Teflon line, with a continuously measured sample of flow of $\sim 12 \text{ L min}^{-1}$. The trace gas inlet used the forward motion of the aircraft to push ambient air into the line in combination with a rear-facing 0.95 cm Teflon exhaust line that lowered the pressure in the sampling line.

2.2 State parameters and winds

State parameters and meteorological conditions were measured with an AIMMS-20, manufactured by Aventech Research Inc. (Barrie, Ontario, Canada; <http://aventech.com/products/aimms20.html>). The AIMMS-20 consists of three modules: (1) an Air Data Probe, which measures temperature and the three-dimensional aircraft-relative flow vector (TAS, angle of attack, and side slip) with a three-dimensional accelerometer for measurement of turbulence; (2) an Inertial Measurement Unit, which provides the aircraft angular rate and acceleration; and (3) a Global Positioning System for aircraft three-dimensional position and inertial velocity. Vertical and horizontal wind speeds are measured with accuracies of 0.75 and 0.50 m s^{-1} respectively. Accuracy and precision of the temperature measurement are 0.30 and $0.10 \text{ }^\circ\text{C}$ respectively.

2.3 Aerosol physical properties

Measurements of particle number concentrations and size were made aboard Polar 6 at a frequency of 1 Hz, unless otherwise indicated. Number concentrations of particles greater than 5 nm in diameter ($N_{>5}$) were measured with a TSI 3787 water-based ultra-fine condensation particle counter (UCPC), sampling at a flow rate of 0.6 L min^{-1} . Aerosol number size distributions from 20 nm to $1 \mu\text{m}$ were acquired with two instruments: a Brechtel Manufacturing Incorporated (BMI) scanning mobility system (SMS) coupled to a TSI 3010 condensation particle counter (CPC) measured from 20 to 100 nm (N_{20-100}) with a 60 s time resolution, while a Droplet Measurement Technology (DMT) Ultra High Sensitivity Aerosol Spectrometer (UHSAS) measured number size distributions from 70 nm to $1 \mu\text{m}$ ($N_{>70}$) with a time resolution of 1 Hz. The SMS sampled at a flow rate of 1 L min^{-1} , with a dried ($\sim 20 \text{ } \%$ RH) sheath flow of 6 L min^{-1} . The UHSAS uses light-scattering signals from a 1054 nm laser for particle detection and sizing on a single-particle basis (e.g., Cai et al., 2008), with a sample flow rate of $55 \text{ cm}^3 \text{ min}^{-1}$ from a bypass flow off the main aerosol inlet. Characterization and calibration of the UCPC, SMS, and UHSAS are described in detail in Leitch et al. (2016). Particle number concentrations from the SMS and UHSAS generally agreed within a factor of 2 over their overlapping size range (70 to 100 nm).

Particle number concentrations from all instruments are reported at ambient pressure and temperature. A character-

istic size distribution is shown in Fig. 1. Values of $N_{>80}$, $N_{>100}$, and $N_{>200}$ were derived from UHSAS measurements. Number concentrations from 5 to 20 nm (N_{5-20}) were estimated by subtracting the sum of the SMS total number concentration (N_{20-100}) and UHSAS $N_{>100}$ from the total UCPC concentration (i.e. $N_{>5}$). The number of particles greater than 50 nm ($N_{>50}$) was determined by the sum of the SMS number from 50 to 100 nm (N_{50-100}) and the UHSAS $N_{>100}$. Here, we refer to N_{5-20} as the nucleation mode, N_{20-100} as the Aitken mode, and $N_{>100}$ and larger as the accumulation mode.

2.4 Cloud condensation nuclei concentrations

CCN concentrations were measured using a DMT CCN counter (CCNC, model 100), sampling behind a DMT pressure-controlled inlet (PCI) at a reduced pressure of $\sim 6.5 \times 10^4$ Pa. The effective supersaturation (for a nominal water supersaturation of 1 %, at 6.5×10^4 Pa) was found to be 0.6 % (Leaith et al., 2016) and was held constant throughout the study to allow more measurement stability and the highest time resolution possible and to examine the hygroscopicity of small particles. Calibration and characterization of the CCNC is described in (Leaith et al., 2016).

The effective aerosol hygroscopicity parameter (κ) was estimated according to Petters and Kreidenweis (2007), using the average aerosol composition from the aerosol mass spectrometer (Sect. 2.6.2) with ammonium sulfate and organic aerosol densities of 1770 and 1550 kg m⁻³ respectively (e.g., Chang et al., 2010). Assuming a temperature of 298 K and the surface tension of pure water the dry diameter for activation was calculated at the supersaturation of our CCN measurements. The measured size distribution could then be integrated down to this dry diameter to produce predicted CCN concentrations for comparison with measured values.

2.5 Trace gases

Carbon monoxide (CO) concentrations were measured with an Aerolaser ultra-fast carbon monoxide monitor (model AL 5002), based on VUV-fluorimetry using excitation of CO at 150 nm. The instrument was modified such that in situ calibrations could be conducted in flight. CO concentrations are used here as a relative indicator of aerosol influenced by pollution sources, such as anthropogenic or biomass burning emissions.

Water vapour (H₂O) measurements were based on infrared absorption using a LI-7200 enclosed CO₂/H₂O Analyzer from LI-COR Biosciences GmbH. In situ calibrations were performed during flight at regular intervals (15–30 min) using a NIST traceable CO₂ standard with zero water vapour concentration. The measurement uncertainty is ± 40 ppmv. H₂O mixing ratios were used to calculate RH with pressure and temperature measured by the AIMMS-20.

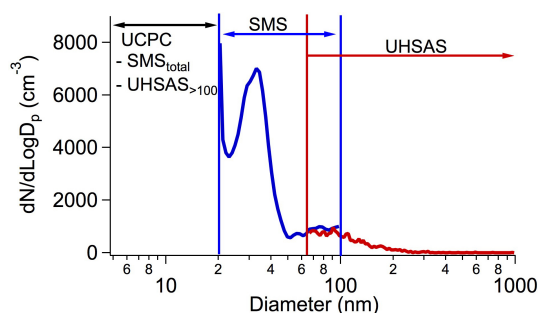


Figure 1. Characteristic size distribution showing the size range of the SMS and UHSAS (observed near 81.1° W in Lancaster sound, see Fig. 2a). Number concentrations from 5 to 20 nm (N_{5-20}) were estimated by subtracting the sum of the SMS total number concentration (N_{20-100}) and UHSAS number concentration greater than 100 nm ($N_{>100}$) from the total UCPC concentration (i.e. $N_{>5}$).

2.6 Sub-micron aerosol composition

2.6.1 Single particle soot photometer

Concentrations of particles containing refractory black carbon (rBC) were measured with a DMT single particle soot photometer (SP2) (described in Schwarz et al., 2006 and Gao et al., 2007) and are used as an indicator of pollution influences. The SP2 uses a continuous intra-cavity Nd:YAG laser (1064 nm) to classify particles as either incandescent (rBC) or scattering (non-rBC), based on the individual particle's interaction with the laser beam. The peak incandescence signal is linearly related to the rBC mass. The SP2 was calibrated with an Aquadag standard by selecting a narrow size distribution of particles with a differential mobility analyzer upstream of the SP2 (Schwarz et al., 2006; Laborde et al., 2012). The detection efficiency of this SP2 (version D) drops off for particles smaller than 70 nm. A log-normal fit through the mass-size distribution indicates that the SP2 measured 92 % of the total ambient rBC mass. Reported rBC values were thus scaled up by a factor of 1.08 to account for the fraction of rBC particles that were outside of the SP2 detection range. The SP2 sampled at 120 cm³ min⁻¹, sharing a bypass line from the main aerosol inlet with the UHSAS.

2.6.2 Aerosol mass spectrometer

Sub-micron aerosol composition was measured with an Aerodyne high-resolution time-of-flight aerosol mass spectrometer (HR-ToF-AMS), described in detail by DeCarlo et al. (2006). The HR-ToF-AMS deployed here was equipped with an infrared laser vaporization module similar to that of the SP2 (DMT); however, measurements of rBC are not relevant for the data presented here due to extremely low rBC concentrations. The HR-ToF-AMS was operated in “V-mode” with a mass range of m/z 3–250, alternating between ensemble mass spectrum (MS) mode for 20 s (two cycles of

5 s MS open and 5 s MS closed) and efficient particle time-of-flight (epToF) mode for 10 s. Filtered ambient air was sampled with the HR-ToF-AMS approximately three times per flight, for a duration of at least 5 min, to account for contributions from air signals. Data were analysed using the Igor Pro based analysis tool PIKA (v.1.16) and SQUIRREL (v.1.57) (Seuper, 2010).

The HR-ToF-AMS sampled behind a PCI system, similar to that described by Hayden et al. (2011), in order to remove variations in particle sizing, transmission, and air-beam signals (used to correct particle signals for variations in instrument sensitivity; Allan et al., 2003) as a function of pressure in the aerodynamic lens (Bahreini et al., 2008; DeCarlo et al., 2008). The PCI system maintained a pressure of 6.19×10^4 Pa upstream of a 130 μm orifice in the HR-ToF-AMS inlet and downstream from a 200 μm orifice, such that the pressure in the aerodynamic lens was maintained at 173 Pa (~ 1.3 Torr) by variable pumping. In this configuration, the lens pressure was adequately maintained up to an altitude of ~ 3500 m. Characterization of particle transmission efficiency with and without the PCI was carried out before and after the study (Sect. 1.1 in the Supplement). Results demonstrated near 100 % transmission of ammonium nitrate particles from ~ 70 to 700 nm (mobility diameter) through the PCI, in comparison to transmission through the aerodynamic lens alone (Fig. S1 in the Supplement). Note that the size range over which AMS lens transmission is optimal can be very instrument dependent. Using the AIMMS measured ambient pressure and the PCI internal pressure, HR-ToF-AMS particulate mass loadings are reported at ambient pressure.

Species comprising non-refractory particulate matter are measured with the HR-ToF-AMS, including sulfate, nitrate, ammonium, and the sum of organic species, with an uncertainty of ± 30 % (Bahreini et al., 2009). The HR-ToF-AMS is capable of detecting other species, including methanesulfonic acid (Phinney et al., 2006; Zorn et al., 2008) and sea salt (Ovadnevaite et al., 2012).

The detection efficiency of sea-salt-containing particles is dependent on not only the ambient RH but also the temperature of the tungsten vaporizer (Ovadnevaite et al., 2012). The vaporizer temperature was calibrated with sodium nitrate particles, such that particle-time-of-flight signals indicated efficient vaporization, and was operated at a temperature of $\sim 650^\circ\text{C}$. HR-ToF-AMS signals for sea salt, in particular NaCl^+ (m/z 57.96), can be used to quantify sea salt mass loadings (e.g., Ovadnevaite et al., 2012); however, here we use the NaCl^+ signal only as a qualitative indication for the presence of sea salt.

After the method of Zorn et al. (2008), we determined the fragmentation pattern for MSA under the operating conditions of our HR-ToF-AMS by utilizing the unique MSA fragment CH_3SO_2^+ (m/z 78.99), which was well resolved from organic fragments at the same nominal mass (i.e. C_6H_7^+ at

m/z 79.05; Fig. S2 in the Supplement). The default HR-ToF-AMS fragmentation table was modified to include MSA, such that contributions from MSA to peaks usually associated with organic species and sulfate were accounted for. The sensitivity of our HR-ToF-AMS to MSA relative to nitrate (RIE_{MSA}) was determined to be 1.33 ± 0.05 , which is similar to estimated values used in other studies (e.g., Zorn et al., 2008). The MSA calibration and fragmentation pattern is described in more detail in Sect. 1.1 of the Supplement.

Ammonium nitrate calibrations (with 300 nm particles) were carried out four times during the campaign (Jimenez et al., 2003), and air-beam corrections were referenced to the appropriate calibration in order to account for differences in instrument sensitivity between flights. The relative ionization efficiencies for sulfate and ammonium (RIE_{SO_4} and RIE_{NH_4}) were 1.4 ± 0.1 and 3.7 ± 0.3 . The default relative ionization efficiency for organic species (i.e. $\text{RIE}_{\text{Org}} = 1.4$) was used (Jimenez et al., 2003), which may lead to some larger uncertainty in the quantification of organic aerosol mass (Murphy, 2015). Elemental composition was calculated using the method presented in Canagaratna et al. (2015). Detection limits for sulfate, nitrate, ammonium, MSA, and organics based on three times the signal-to-noise ratio of filter measurements in flight were 0.009, 0.008, 0.004, 0.005, and $0.08 \mu\text{g m}^{-3}$ respectively. A composition-dependent collection efficiency (CDCE) was applied to correct HR-ToF-AMS mass loadings for non-unity particle detection due to particle bounce on the tungsten vaporizer (Middlebrook et al., 2012). After the CDCE correction HR-ToF-AMS, total mass loadings agreed with estimated mass concentrations from the UHSAS within a factor of 2.

2.6.3 Aircraft-based Laser Ablation Aerosol Mass Spectrometer (ALABAMA)

Single particle analysis was conducted using the ALABAMA. A detailed description of the instrument can be found in Brands et al. (2011). Briefly, the ALABAMA samples particles through a PCI and an aerodynamic lens. The particles are detected and sized by light scattering when passing two continuous laser beams separated along the path of the sampled aerosol. Particles are ablated and ionized by a single 266 nm laser pulse, and the resulting ions are detected in a bipolar time-of-flight mass spectrometer. Optical detection of aerosol limits the minimum detectable particle size to approximately 150 nm with particles at approximately 400 nm detected at the highest efficiency. The transmission efficiency in the aerodynamic lens limits the maximum detectable size to approximately 1000 nm. Particle mass spectra collected by the ALABAMA are analysed using a software package that includes m/z calibration, peak area integration, and automated clustering using fuzzy c-means clustering (Hinz et al., 1999; Roth et al., 2016). As is done in this case study, subsets of particles can also be analysed manu-

ally by searching for selected marker peaks known from reference laboratory and field data.

A subset of 68 particles detected over the period relevant to this case study was analysed manually using marker peaks as follows. Organic carbon (OC) was characterized by peaks at m/z 27, 37 and 43 ($C_2H_3^+$, C_3H^+ , and CH_3CO^+ or $C_3H_7^+$). Pronounced peaks at m/z multiples of 12 (e.g., 12, 24, ..., 108) ($C_n^{+/-}$) identify elemental carbon (EC). Mass spectra containing peaks at m/z multiples of 12, but not higher than 36, can be either fragments of elemental or organic carbon and are therefore designated here as EC/OC. MSA was identified by a peak at m/z 95 ($CH_3SO_3^-$). Interference from $Na^{37}Cl_2^-$ is unlikely if no m/z 93 ($Na^{35}Cl_2^+$) is present. Further marker peaks include m/z 97 for sulfate (S), m/z 26 and 42 for CN^- and CNO^- (CN), m/z 39 and 41 (K^+) for potassium (K), and m/z 40, 56, and 57 (Ca^+ , CaO^+ , $CaOH^+$) for calcium (Ca). The presence of sodium chloride (NaCl) was determined by peaks at m/z 23, 35, 37, 81, and 83 (Na^+ , Cl^- , and Na_2Cl^+). Due to chemical aging processes, Cl^- can be replaced by nitrate, resulting in the presence of peaks at m/z 46 and 62 (NO_2^- and NO_3^-) in addition to sodium chloride. Trimethylamine (TMA) was identified by peaks at m/z 58 and 59 ($C_3H_8N^+$ and $C_3H_9N^+$) based on laboratory reference measurements of TMA particles and previously published field data (e.g., Rehbein et al., 2011; Healy et al., 2015).

2.7 Identifying air mass history using the FLEXible PARTicle dispersion model driven by meteorology from the Weather Research and Forecasting model (FLEXPART-WRF)

The Lagrangian particle dispersion model FLEXPART-WRF (Brioude et al., 2013) (website: <http://flexpart.eu/wiki/FpLimitedareaWrf>) was used to study the history air masses prior to sampling during the flight. FLEXPART-WRF is based on FLEXPART (Stohl et al., 2005) but uses the limited area meteorological forecast from WRF (Skamarock et al., 2001), with the specific WRF forecast details for the NETCARE campaign provided in Wentworth et al. (2016). Here, we use FLEXPART-WRF run in backward mode to study the origin of air influencing aircraft-based aerosol measurements. Further details of the FLEXPART-WRF simulations performed for NETCARE 2014 summer campaign are also found in Wentworth et al. (2016).

3 Results and discussion

3.1 Flight overview and meteorological situation

In this case study we focus on the flight conducted on 12 July 2014 where Polar 6 travelled at ~ 3 km altitude from Resolute Bay, past the marginal ice zone and out over open water to the eastern end of Lancaster Sound (Fig. 2a) as far as was

permitted by our aircraft range, at which point it descended and returned west. The relevant portion of this flight, over open water in Lancaster Sound, is highlighted in Fig. 2a (79.7 to 86.5° W, 20:00–21:20 UTC). During this time the aircraft flew to the west below 100 m a.g.l. and covered a distance of approximately 175 km at a survey speed of 75 ms^{-1} under clear sky conditions. Profiles were carried out at three different locations to characterize the vertical structure of the troposphere (Fig. 2a, triangles): one profile from 60 to 3000 m near Resolute Bay ($\sim 95^\circ$ W, denoted as “west” in Fig. 2b and c); a second, shallower profile to ~ 900 m near the marginal ice zone ($\sim 88^\circ$ W, “central”); and a third profile down from 3000 to 60 m in eastern Lancaster Sound ($\sim 80^\circ$ W, “east”).

Meteorological observations and measurements of trace gases and black carbon indicate a stable and clean atmosphere. Temperature profiles in all three locations indicated a shallow surface-based temperature inversion of $2\text{--}4^\circ\text{C}$, reaching up to ~ 800 m over the ice near Resolute Bay and to only ~ 100 m in the eastern profile (Fig. 2b). Applying the method of bulk Richardson number (Aliabadi et al., 2016a) with Polar 6 meteorological observations, and radiosondes conducted concurrently at Resolute Bay and aboard the CCGS Amundsen, Aliabadi et al. (2016b) estimated boundary layer heights of 275 ± 164 m during the NETCARE summer campaign. Here, we will refer to the portion of the boundary layer with a positive vertical gradient in the temperature profile as the “lower boundary layer.” Both within and above the lower boundary layer winds were predominantly from the west, with measured wind speeds near the surface averaging $6.5 \pm 1.8 \text{ ms}^{-1}$. Similarly, surface winds from WRF indicate predominately west-north-west winds at this time, with wind speeds of $4\text{--}8 \text{ ms}^{-1}$ (Fig. S4 in the Supplement). CO profiles in all three locations demonstrated very clean background conditions with concentrations ranging from 73 to 78 ppbv and little variation with altitude (Fig. 2c). RH was generally high near the surface, with an average of 80 % in the lower boundary layer (Fig. S5 in the Supplement). Refractory black carbon concentrations also indicate a very clean atmosphere, well below the threshold for a clean marine boundary layer discussed by Gantt and Meskhidze (2013). Average (\pm standard deviation) rBC mass loadings during the period of interest and over the entire flight were 1.6 ± 0.9 and $2.5 \pm 1.5 \text{ ng m}^{-3}$ respectively, with slightly higher concentrations found aloft.

Air mass history from FLEXPART-WRF indicates a strong local Arctic influence on the sampled air mass. FLEXPART-WRF air mass origin is shown as the column-integrated air mass residence time prior to sampling, also referred to as the column-integrated potential emission sensitivity (PES), for the release time and location of this case study (Fig. 3a). The column-integrated PES supports that the locally influenced air mass originated from generally clean conditions with no pollution sources. The air mass encountered by the aircraft at 82.2° W and ~ 85 m a.g.l. resided over

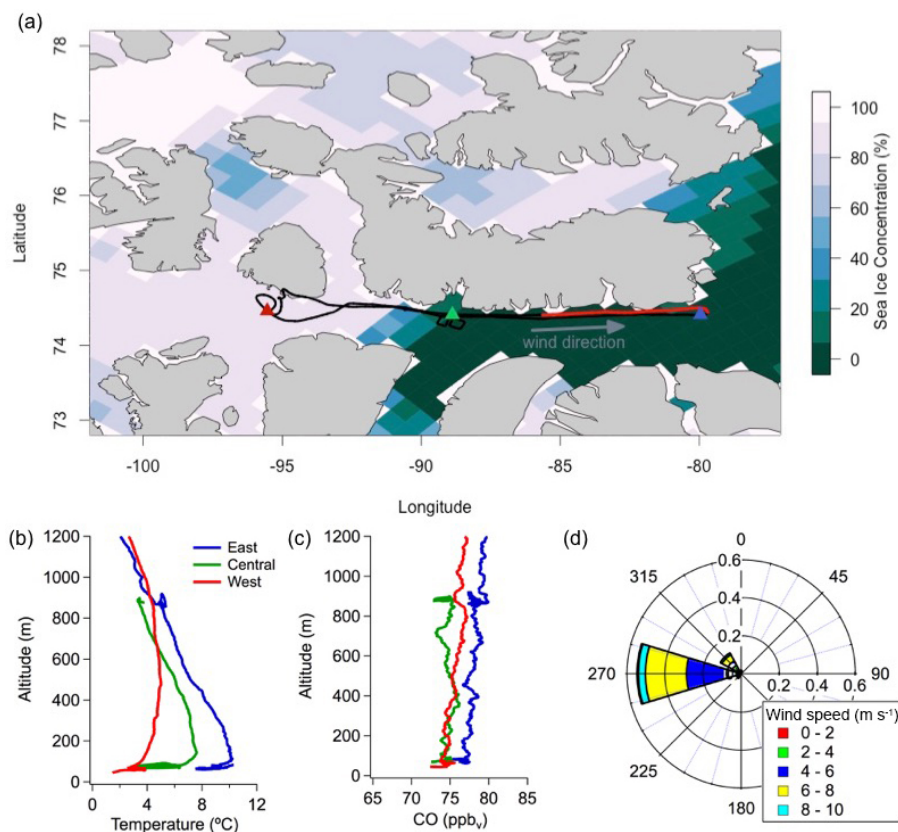


Figure 2. (a) Map of the study area showing sea ice concentration for 12 July 2014 from the National Snow and Ice Data Center (<http://nsidc.org>; Cavalieri et al., 1996) and the flight track originating at Resolute Bay, Nunavut ($74^{\circ}41'N$, $94^{\circ}52'W$), and extending to eastern Lancaster Sound. The case study area is highlighted in red (20:00–21:20 UTC, 79.7 to $86.5^{\circ}W$), at which time the aircraft travelled westward at ~ 100 m a.g.l. The prevailing wind direction is marked with an arrow. Triangles mark the location at which the aircraft reached ~ 1 km a.g.l. during each profile shown in panels (b) and (c). (b, c) Profiles of temperature and CO mixing ratio near Resolute Bay (red), in central Lancaster Sound (green), and in eastern Lancaster Sound (blue). (d) Flight-average wind rose; wind speeds at the surface averaged ~ 6.5 m s⁻¹.

a snow and ice-covered island (Devon Island; Friedl et al., 2010) for approximately 1 week before descending into Lancaster Sound within 1 day of sampling (Fig. 3). The model also indicates that the sampled air mass had a residence time within the lowest 300 m of 4 to 5 h prior to sampling, providing at least 4 h of transport and chemistry within the boundary layer (Fig. 3b and c). Overall, FLEXPART-WRF air mass history suggests that the sampled air mass had little exposure to fresh sea emissions until 4 to 5 h prior to sampling, when it moved from above the snow- and ice-covered land and was exposed to the ocean surface within the lower boundary layer.

3.2 Observations of particle growth

Our observations of particle number concentration, over the size range from 5 nm to 1 μ m, indicated the simultaneous presence of nucleation mode and Aitken mode particles near the ocean surface. At low altitude near $85^{\circ}W$ we observed an enhancement in N_{5-20} above background levels, indicating the presence of nucleation mode particles (Figs. 1, 4a).

Upon entering the lower boundary layer further downwind (Fig. 2d) and to the east (82.5 – $81^{\circ}W$), we observed a sharp increase in N_{5-20} concurrently with an increase in N_{20-100} (Fig. 4a). We do not directly observe the formation of the smallest particles; however, we hypothesize that they were formed through nucleation in a very clean atmosphere.

Particle number size distributions from 20 to 1000 nm illustrate that particles below 20 nm (Fig. 4a, c) grow to form a mode centred at 30–40 nm (Fig. 4d–f). Beyond $86^{\circ}W$ we observe N_{20-100} at background levels of ~ 100 cm⁻³. These observations suggest that the aerosol size distribution develops as the air mass moves downwind to the east. The advection timescale from 85.8 to $81.1^{\circ}W$ is 6.7 h, given an average wind speed of 6.5 m s⁻¹, and the sampling time of the aircraft over this distance is 35 min. Given the substantial changes in aerosol size and number concentration observed over this relatively short time period, our observations suggest that a source of condensable material contributing to aerosol growth is present to the west of the sampling region and it is unlikely that a wider source region contributed. An estimate of the

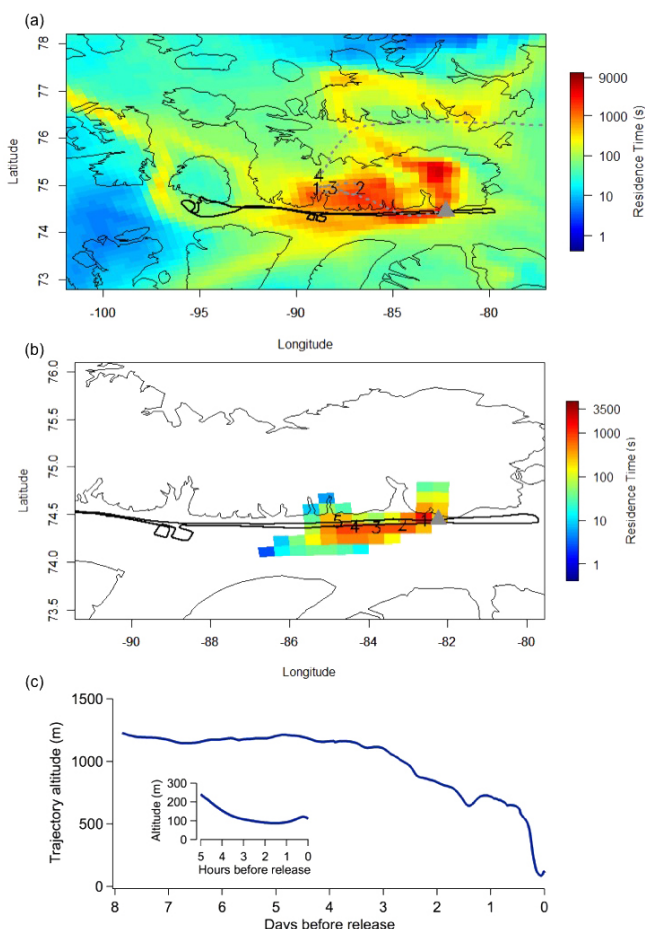


Figure 3. (a) Total column air mass residence time predicted by FLEXPART-WRF, indicating the origin of air sampled along the flight track. The aircraft location at the time of the FLEXPART-WRF particle release is indicated with a grey triangle (74.4° N, 82.2° W; ~85 m a.g.l.; 20:39:25 UTC). The colour scale represents the residence time of air, in seconds, at a particular location before arriving at the aircraft position. The plume centroid location is shown with a grey dashed line. Numbers indicate the plume centroid location, in days prior to release. (b) Partial column (below 300 m) PES predicted by FLEXPART-WRF shown as residence time in seconds for particles released at the aircraft location in panel (a). The colour scale shows the residence time of particles for 5 h prior to the release time and below 300 m. Numbers indicate the plume centroid location, in hours prior to release. (c) Plume centroid altitude 8 days prior to release and 5 h prior to release (inset).

growth rate in this case is associated with a large uncertainty since it is complicated by a number of factors, including the 1 min time resolution of the SMS that corresponds to a sampling distance of ~4 km, and uncertainties in the advection time. Therefore, it is difficult to quantitatively follow the evolution of the size distribution. Compounded by our lack of knowledge of the spatial uniformity of the condensable material, we do not present an estimate of the growth rate here.

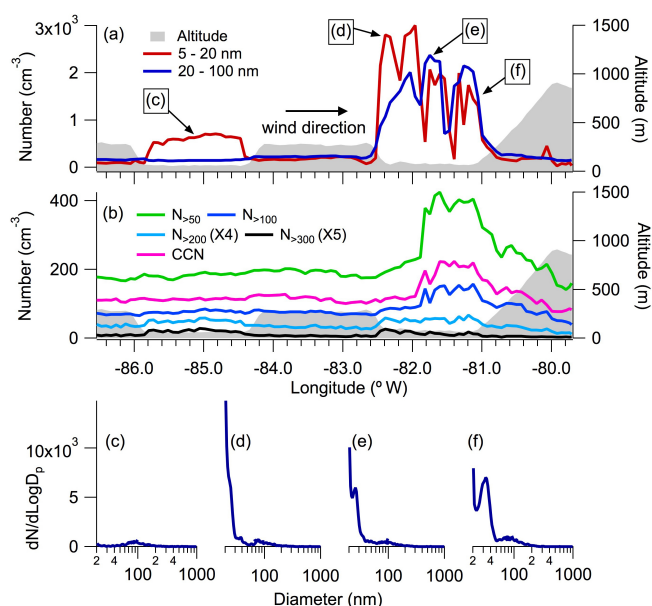


Figure 4. (a) Aircraft altitude (grey) and particle number concentrations from 5–20 nm (N_{5-20} , red) and 20–100 nm (integrated SMS concentration, N_{20-100} , dark blue), both shown at the time resolution of the SMS, over the case study area highlighted in Fig. 2a. (b) Particle number concentrations greater than 50 nm ($N_{>50}$, light green), greater than 100 nm ($N_{>100}$, dark blue), greater than 200 nm ($N_{>200}$, light blue, multiplied by 4), greater than 300 nm ($N_{>300}$, black, multiplied by 5) and CCN concentrations at 0.6 % supersaturation (pink) shown at the time resolution of the SMS. Particle-number size distributions from 20 to 1000 nm (from the SMS and UHSAS) at (c) 85.1° W, (d) 82.3° W, (e) 81.8° W and (f) 81.1° W.

The boundary layer was characterized by a low pre-existing aerosol surface area (i.e. a small condensation sink). A small number of particles above 200 nm in diameter (~10–15 cm⁻³) were present within the lower boundary layer and show a time variation distinct from that of $N_{>50}$ and $N_{>100}$ (Fig. 4b). These larger particles are present during both sampling periods within the lower boundary layer (i.e. near 85 and 82.5° W), where winds speeds were relatively constant (6.5 ± 1.8 m s⁻¹), and could be from ejection of primary sea-spray aerosol (see Sect. 3.3.2). The small $N_{>200}$ provides a low pre-existing aerosol surface area (average \pm standard deviation: 3.8 ± 2.0 μ m² cm⁻³), which assists particle nucleation.

$N_{>50}$ and $N_{>100}$ show a time variation distinct from that of the nucleation mode (N_{5-20}) and larger accumulation mode particles ($N_{>200}$) (Fig. 4b). In our eastern-most observations within the lower boundary layer (82.5–81° W), $N_{>50}$, which is accounted for largely by N_{50-150} , is enhanced above background levels of ~200 to ~400 cm⁻³ (Fig. 4b). At the same time, CCN concentrations are elevated to > 200 cm⁻³, above background levels of ~100 cm⁻³ (Fig. 4b). $N_{>50}$, $N_{>100}$, and CCN concentrations remain somewhat elevated up to ~900 m (Fig. 4b, near 80.5° W), suggesting that some mix-

ing above the lower boundary layer occurred during some time prior to our observations, possibly due to katabatic winds off Devon Island suggested from the FLEXPART-WRF analyses (Fig. 3c). Profiles of aerosol number and composition are presented in Fig. S5 in the Supplement.

The variation in the size distribution from west to east in Lancaster Sound suggests that particles between ~ 30 and greater than ~ 50 nm are forming through secondary processes. In our western-most observations in the lower boundary layer (85.6 – 84.4° W) the size distribution is dominated by particles below 20 nm (Fig. 4a, c). Further to the east in the lower boundary layer (82.5 – 81° W), N_{5-20} is high and subsequently decreases moving east (Fig. 4a), while $N_{>50}$ and $N_{>100}$ do not increase until 81.9° W (Fig. 4b). The aircraft covered a distance of 19 km between entering the lower boundary layer (82.5° W) and observing this increase in $N_{>50}$ and $N_{>100}$. With a wind speed of 6.5 ms^{-1} near the surface, the advection time over this distance is approximately 50 min. If the aerosol size distribution was dominated by primary sea-spray aerosol, given constant wind speed, there would be no reason for such a delay in our observations of $N_{>50}$ and $N_{>100}$. Indeed, given the decreasing abundance of $N_{>300}$, the evidence suggests that the sea-spray source, which is associated with larger particles (e.g., Modini et al., 2015, see Sect. 3.3.2), is becoming less important as $N_{>50}$ is increasing. These observations are suggestive of a secondary process growing particles from less than 20 nm into larger sizes, above 50 nm.

3.3 Aerosol composition

3.3.1 Carbonaceous aerosol

Our observations of particle growth are correlated with an increase in OA and MSA in sub-micron particles (Fig. 5a), corresponding to increased organic-to-sulfate and MSA-to-sulfate ratios (Fig. 5b). The presence of MSA, an intermediate-volatility oxidation product of DMS, indicates a marine-biogenic influence on the aerosol sulfur (Bates et al., 1992). MSA cannot be viewed as a conservative tracer of DMS oxidation (e.g., Bates et al., 1992); however, it is notable that the MSA-to-sulfate ratio reached a peak value of 0.15 during the growth event (corresponding to a peak mass of 60 ng m^{-3}), which is significantly higher than at all other times during this flight (Fig. 5b). The absolute MSA concentration measured by the HR-ToF-AMS should be viewed as a lower limit since a portion of the MSA mass could reside on particles smaller than the lower size limit of the instrument. Particle-size-resolved mass spectra (pToF, Fig. S6 in the Supplement) during particle growth indicate that total organic aerosol was present in relatively small particle sizes, from less than 80 nm to approximately 200 nm (vacuum aerodynamic diameter, d_{va}). Unfortunately, signal-to-noise ratios for MSA were such that little useful information could be drawn from the corresponding pToF data. The correlation of

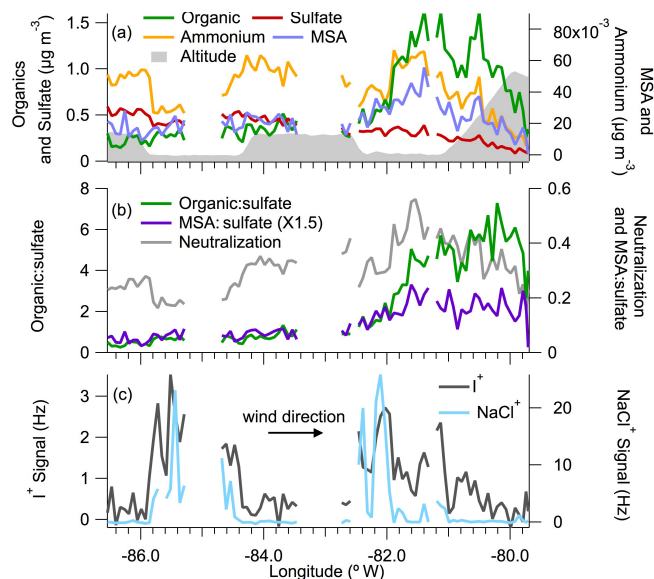


Figure 5. (a) Organic species, sulfate, ammonium, and methane-sulfonic acid (MSA) measured by the HR-ToF-AMS, over the case study area highlighted in Fig. 2a. Altitude is shown in grey on the same scale as Fig. 4. (b) Organic-to-sulfate ratio (green), MSA-to-sulfate ratio (purple), and extent of neutralization (grey). The extent of neutralization is the ratio of measured to predicted ammonium, based on measured sulfate, nitrate, and MSA. (c) I^+ (m/z 126.90) and NaCl^+ signal from the HR-ToF-AMS.

OA and MSA with particle growth suggests that the growth of particles into the size range of the HR-ToF-AMS was mediated by the condensation of MSA and condensible organic species. The source and identity of these species, aside from MSA, is not known, but we hypothesize a role for marine-derived biogenic volatile organic compounds (VOCs).

While non-marine sources of condensible organic species, such as emissions of isoprene and terpenes from high Arctic terrestrial vegetation (Schollert et al., 2014) and photochemical production of VOCs in the snowpack (Grannas et al., 2007) (e.g., over the snow- and ice-covered Devon Island), could also contribute to particle growth, single particle observations of aerosol composition further suggest a marine influence on particles greater than ~ 150 nm (d_{va}). Fifty-four percent of particles detected by the ALABAMA over the region highlighted in Fig. 2a contained detectable signal for TMA (Fig. 6), in support of aerosol growth through the condensation of marine-derived biogenic VOCs (e.g., Facchini et al., 2008a; Dall’Osto et al., 2012). Consistent with HR-ToF-AMS observations of MSA during the growth event, $\sim 30\%$ of particles detected by the ALABAMA contained observable MSA signal. TMA was mainly present as an internal mixture with potassium, sulfate, other organic species, and to a lesser degree with MSA (Fig. 6).

Organic aerosol observed by the HR-ToF-AMS during particle growth appears chemically distinct from the OA ob-

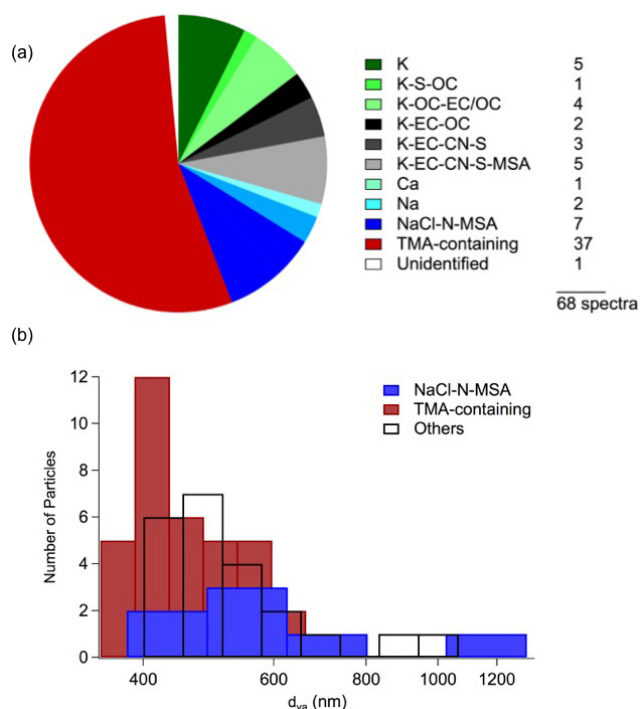


Figure 6. (a) Pie chart depicting particle types detected by the AL-ABAMA over the case study area highlighted in Fig. 2a. Particles were grouped based on the presence of marker peaks and the particle group name indicates the relative abundance of the corresponding signals in particle spectra. A total of 68 particle spectra were obtained during the approximately 2 h period. TMA-containing particles are mostly internally mixed with K, S, OC, and to a lesser degree MSA and EC / OC. Not all TMA-containing particles included signal for MSA; 13 % of all detected particles contained both TMA and MSA signals. (b) Size distributions (in terms of vacuum aerodynamic diameter, d_{va}) of TMA-containing particles (red), NaCl-containing particles (blue), and all other particles classes (transparent).

served at other times during this flight, especially compared to that above the lower boundary layer (OA mass spectra are presented in Fig. S7 of the Supplement). Hydrocarbon fragments ($C_xH_y^+$, largely unsaturated) contribute 50 % to growth event OA mass spectra and only 30 % to non-growth event OA. Oxygenated organic fragments ($C_xH_yO_z^+$) contribute 50 % to growth event OA mass spectra and 70 % to non-growth event OA. $C_xH_y^+$ and $C_xH_yO_z^+$ fragments are correlated during the growth event, suggesting that these less-oxygenated and more-oxygenated species are arising from a similar source. Average elemental composition also shows notable differences with oxygen-to-carbon (O : C) and hydrogen-to-carbon (H : C) ratios in the growth event OA of 0.5 and 1.6, while non-growth event OA was significantly more oxygenated with O : C and H : C ratios of 0.78 and 1.2, suggesting less aged OA during the growth event compared to other times.

To gain further insight into the characteristics of the OA observed during the growth event, we compared our MS with a number of OA mass spectra obtained with AMS instruments. The growth event OA compares favourably with marine-like OA observed at Mace Head, Ireland ($R^2 = 0.75$) (Ovadnevaite et al., 2011), as well as with marine OA observed over the Arctic Ocean ($R^2 = 0.88$) (Chang et al., 2011a). OA from the growth event also compares favourably with α -pinene secondary organic aerosol (SOA) generated under low NO_x conditions ($R^2 = 0.78$) (Chhabra et al., 2011) and with spectra associated with isoprene SOA from a forested site ($R^2 = 0.85$) (Robinson et al., 2011), but it does not compare well with IEPOX SOA ($R^2 = 0.07$) (Bougiatioti et al., 2013). In conjunction with the presence of MSA during the growth event, the comparisons with previously observed marine-OA spectra support the hypothesis that we observe a marine-influenced aerosol. The comparisons with terpene-related OA could support a marine-influenced aerosol (e.g., S. Shaw et al., 2010) but could also be consistent with other regional sources of these OA precursors (e.g., Grannas et al., 2007; Schollert et al., 2014).

3.3.2 Other aerosol chemical species

Other aerosol components detected by the HR-ToF-AMS showed a time variation distinct from organic aerosol species. Sulfate mass loading was relatively constant, within the lower boundary layer (Fig. 5a), suggesting that it did not contribute significantly to particle growth during this event. Due to the relatively slower oxidation of sulfur dioxide to sulfuric acid, it is feasible that MSA resulting from DMS oxidation could be contributing to particle growth while sulfate salts are not. However, this would be inconsistent with the results of Giamarelou et al. (2016). Similarly to the observed OA, sulfate was present in relatively small particles with a peak in the size distribution slightly larger than that of OA (Fig. S6 in the Supplement). Ammonium concentrations are low and show some correlation with organic and inorganic aerosol species, suggesting that OA, MSA, and sulfate could be partially neutralized by ammonium. The HR-ToF-AMS estimate of aerosol neutralization (accounting for sulfate, nitrate, and MSA) peaks at a value of ~ 0.6 during particle growth (Fig. 5b).

Exclusively within the lower boundary layer we observe an increase in iodine signal as I^+ (m/z 126.90), while no other iodine-containing peaks were observed above mass spectral noise (Fig. 5c). Our observations are potentially consistent with those of Allan et al. (2015), who used similar measurements to highlight the possible role of iodine-oxide species in particle nucleation in Arctic regions. Here, I^+ shows a modest correlation not only with N_{5-20} but also with $N_{>200}$, since particles in both size ranges are confined to the lower boundary layer and their variability in time is largely dictated by the aircraft's position (Fig. S8 in the Supplement). Without further information about the chemical form

of the iodine we observe, it is difficult to discern whether the HR-ToF-AMS I^+ arises from iodine-oxides present in small particles or from biological iodine-containing compounds and iodine-containing salts potentially present in primary sea-spray aerosol (e.g., Murphy et al., 1997).

Primary sea-spray aerosol was confined to the lower boundary layer and contributed largely to $N_{>200}$. The HR-ToF-AMS signal for $NaCl^+$, qualitatively indicating the presence of sea salt aerosol, is present in the lower boundary layer (Figs. 5c and S5 in the Supplement) and correlates well with $N_{>200}$ and $N_{>300}$ (Fig. S8 in the Supplement). This result is supported by Modini et al. (2015), who found a primary marine aerosol mode peaking near 200 nm in ambient observations of marine aerosol off the coast of California. As mentioned above, the negative relationship between $N_{>50}$ and both $N_{>300}$ and $NaCl^+$ near 82.5° W suggests a decreasing importance of primary sea spray at the point where the secondary formation is maximum. Consistent with this observation, single particle measurements from the ALABAMA indicate that $NaCl$ -containing particles were present at larger sizes (i.e. peaking at 400 nm d_{va}) and, notably, were externally mixed from other particle types containing TMA (Fig. 6).

3.4 Cloud condensation nuclei

CCN concentrations are elevated above background levels during the growth event and are well-correlated with the number of particles greater than 80 nm ($N_{>80}$, Fig. 7). If the particles contributing to CCN concentrations at this time were only composed of ammonium sulfate, under our experimental conditions (i.e. 0.6 % supersaturation), we would expect the CCN-activation diameter to be ~ 40 nm (Petters and Kreidenweis, 2007). A CCN-activation diameter of approximately 80 nm therefore indicates that a species less hygroscopic than ammonium sulfate is contributing to the CCN we observe. This is consistent with the elevated OA mass loading we measure when CCN concentrations are high (Fig. 7, colour scale), while sulfate was relatively low compared to other time periods (Fig. 7, marker size).

Since the aerosol was not actively dried and the supersaturation was held constant in the CCNC, in order to allow for rapid measurements, a calculation of the effective aerosol hygroscopicity parameter (κ) in this case carries a large uncertainty (Petters and Kreidenweis, 2007). In particular, measured particle diameters may be slightly larger than the corresponding dry diameter. The temperature in the inlet line was 10–15 °C warmer than the ambient temperature so that the RH decreased significantly as the aerosol entered the sampling line (i.e. during the case study period, the ambient RH was 80 % at 8–10 °C, and the RH decreased within the inlet to approximately < 30 %). Using the measured aerosol composition, we estimate that measured particle diameters are up to 10 % larger than the corresponding dry diameter.

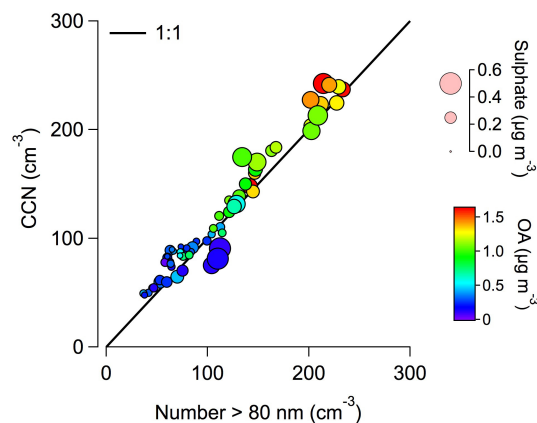


Figure 7. Correlation between the number of particles greater than 80 nm ($N_{>80}$, measured by the UHSAS) and the cloud condensation nuclei (CCN) concentration at 0.6 % supersaturation, below 1 km, during the case study period. Data are coloured by organic aerosol loading and point size corresponds to sulfate loading.

Nonetheless, this calculation is still illustrative of the organic aerosol properties in this environment. If the κ value of the organic aerosol (κ_{Org}) is 0.1, and κ for the whole aerosol is calculated based on the HR-ToF-AMS organic and sulfate loadings (average organic volume fraction = 0.85 ± 0.03) and the known κ for ammonium sulfate, then the resulting dry diameter for activation is ~ 60 nm. From our measurements, the activation diameter seems to be larger than 60 nm so that κ_{Org} of 0.1 could be regarded as an upper limit. When we overestimate aerosol size by 10 %, due to incomplete drying in our sampling line, then our estimated CCN-activation diameter and the calculated dry diameter for activation become more similar. Overall, this illustrates that the organic aerosol was relatively non-hygroscopic with $\kappa_{Org} \sim 0.1$. This estimate is within the range of κ_{Org} recently measured in a coastal, marine-influenced environment by Yakobi-Hancock et al. (2014).

4 Conclusions

In this case study, we present evidence that growth of nucleation mode particles in the summertime Arctic can be mediated by the condensation of MSA and condensible organic species. Our observations of particle growth, informed by observations of particle composition, suggest a combination of primary and secondary aerosol across the size distribution. We observe the growth of small particles, less than 20 nm, into sizes above 50 nm, while our measurements suggest that ejection of primary sea-spray aerosol contributes to externally mixed particles larger than 200 nm. The small $N_{>200}$, which are likely from direct emissions of sea spray, could contain a substantial fraction of OA. However, the majority of OA mass observed here is best correlated with MSA, $N_{>80}$ (dominated by N_{80-150}), and the presence of TMA, suggest-

ing that this OA is largely secondary in origin. In addition, it occurs simultaneously with a period of pronounced aerosol growth. Together, this indicates that the CCN we observe are largely controlled by secondary processes.

Very few studies have measured aerosol composition at high time resolution in the summertime Arctic. Even fewer studies have provided evidence for secondary organic aerosol formation in Arctic regions, in part owing to the infrequency of measurements in the remote marine boundary layer. These results highlight the potential importance of secondary marine organic aerosol formation and its role in growing nucleation mode particles into CCN-active sizes in the clean summertime Arctic atmosphere. Future measurements of nucleation and Aitken mode particle composition coupled to characterization of gas-phase organic species will greatly improve our understanding of particle formation and growth in remote regions, aiding in our ability to understand resulting aerosol–cloud–climate interactions.

5 Data availability

NETCARE (Network on Climate and Aerosols, 2015, <http://www.netcare-project.ca>), which organized the aircraft flight described in this paper, is moving towards a publicly available, online data archive. In the meantime, the data can be accessed by contacting the principal investigator of the network: Jon Abbatt at the University of Toronto (jabbatt@chem.utoronto.ca).

Two external data sets were used in this work and are referenced in the main text. Sea ice concentrations from Nimbus-7 SMMR and DMSP SSM/ISSMIS passive microwave data are available at http://nsidc.org/data/docs/daac/nsidc0051_gsfc_seaice.gd.html (Cavalieri et al., 1996), and MODIS Land cover data are available at <http://glcf.umd.edu/data/lc/> (Friedl et al., 2010).

The Supplement related to this article is available online at doi:10.5194/acp-16-7663-2016-supplement.

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References

- Aliabadi, A. A., Staebler, R., de Grandpré, J., Zadra, A., and Vaillancourt, P.: Comparison of estimated atmospheric boundary layer mixing height in the Arctic and Southern Great Plains under statically stable conditions: Experimental and numerical aspects, *Atmos. Ocean*, 54, 60–74, doi:10.1080/07055900.2015.1119100, 2016a.
- Aliabadi, A. A., Staebler, R., Liu, M., and Herber, A.: Characterization and parameterization of Reynolds stress and turbulent heat flux in the stably-stratified lower Arctic troposphere using aircraft measurements, *Bound.-Lay. Meteorol.*, doi:10.1007/s10546-016-0164-7, 2016b.
- Allan, J. D., Alfarra, M. R., Bower, K. N., Williams, P. I., Gallagher, M. W., Jimenez, J. L., McDonald, A. G., Nemitz, E., Canagaratna, M. R., Jayne, J. T., Coe, H., and Worsnop, D. R.: Quantitative sampling using an Aerodyne aerosol mass spectrometer 2. Measurements of fine particulate chemical composition in two U.K. cities, *J. Geophys. Res.-Atmos.*, 108, 4091, doi:10.1029/2002JD002359, 2003.
- Allan, J. D., Williams, P. I., Najera, J., Whitehead, J. D., Flynn, M. J., Taylor, J. W., Liu, D., Darbyshire, E., Carpenter, L. J., Chance, R., Andrews, S. J., Hackenberg, S. C., and McFiggans, G.: Iodine observed in new particle formation events in the Arctic atmosphere during ACCACIA, *Atmos. Chem. Phys.*, 15, 5599–5609, doi:10.5194/acp-15-5599-2015, 2015.
- Almeida, J., Schobesberger, S., Kurten, A., Ortega, I. K., Kupiainen-Maatta, O., Praplan, A. P., Adamov, A., Amorim, A., Bianchi, F., Breitenlechner, M., David, A., Dommen, J., Donahue, N. M., Downard, A., Dunne, E., Duplissy, J., Ehrhart, S., Flagan, R. C., Franchin, A., Guida, R., Hakala, J., Hansel, A., Heinritzi, M., Henschel, H., Jokinen, T., Junninen, H., Kajos, M., Kangasluoma, J., Keskinen, H., Kupe, A., Kurten, T., Kvashin, A. N., Laaksonen, A., Lehtipalo, K., Leiminger, M., Leppa, J., Loukonen, V., Makhmutov, V., Mathot, S., McGrath, M. J., Nieminen, T., Olenius, T., Onnela, A., Petaja, T., Riccobono, F., Riipinen, I., Rissanen, M., Rondo, L., Ruuskanen,

- T., Santos, F. D., Sarnela, N., Schallhart, S., Schnitzhofer, R., Seinfeld, J. H., Simon, M., Sipila, M., Stozhkov, Y., Stratmann, F., Tome, A., Trostl, J., Tsagkogeorgas, G., Vaattovaara, P., Viisanen, Y., Virtanen, A., Vrtala, A., Wagner, P. E., Weingartner, E., Wex, H., Williamson, C., Wimmer, D., Ye, P., Yli-Juuti, T., Carslaw, K. S., Kulmala, M., Curtius, J., Baltensperger, U., Worsnop, D. R., Vehkamäki, H., and Kirkby, J.: Molecular understanding of sulphuric acid-amine particle nucleation in the atmosphere, *Nature*, 502, 359–363, doi:10.1038/nature12663, 2013.
- Bahreini, R., Dunlea, E. J., Matthew, B. M., Simons, C., Docherty, K. S., DeCarlo, P. F., Jimenez, J. L., Brock, C. A., and Middlebrook, A. M.: Design and Operation of a Pressure-Controlled Inlet for Airborne Sampling with an Aerodynamic Aerosol Lens, *Aerosol Sci. Tech.*, 42, 465–471, doi:10.1080/02786820802178514, 2008.
- Bahreini, R., Ervens, B., Middlebrook, A. M., Warneke, C., de Gouw, J. A., DeCarlo, P. F., Jimenez, J. L., Brock, C. A., Neuman, J. A., Ryerson, T. B., Stark, H., Atlas, E., Brioude, J., Fried, A., Holloway, J. S., Peischl, J., Richter, D., Walega, J., Weibring, P., Wollny, A. G., and Fehsenfeld, F. C.: Organic aerosol formation in urban and industrial plumes near Houston and Dallas, Texas, *J. Geophys. Res.-Atmos.*, 114, D00F16, doi:10.1029/2008JD011493, 2009.
- Bates, T. S., Calhoun, J. A., and Quinn, P. K.: Variations in the methanesulfonate to sulfate molar ratio in submicrometer marine aerosol particles over the south Pacific Ocean, *J. Geophys. Res.-Atmos.*, 97, 9859–9865, doi:10.1029/92JD00411, 1992.
- Bigg, E. K. and Leck, C.: Cloud-active particles over the central Arctic Ocean, *J. Geophys. Res.-Atmos.*, 106, 32155–32166, doi:10.1029/1999JD901152, 2001.
- Bougiatioti, A., Zampas, P., Koulouri, E., Antoniou, M., Theodosi, C., Kouvarakis, G., Saarikoski, S., Mäkelä, T., Hillamo, R., and Mihalopoulos, N.: Organic, elemental and water-soluble organic carbon in size segregated aerosols, in the marine boundary layer of the Eastern Mediterranean, *Atmos. Environ.*, 64, 251–262, doi:10.1016/j.atmosenv.2012.09.071, 2013.
- Brands, M., Kamphus, M., Böttger, T., Schneider, J., Drewnick, F., Roth, A., Curtius, J., Voigt, C., Borbon, A., Beekmann, M., Bourdon, A., Perrin, T., and Borrmann, S.: Characterization of a Newly Developed Aircraft-Based Laser Ablation Aerosol Mass Spectrometer (ALABAMA) and First Field Deployment in Urban Pollution Plumes over Paris During MEGAPOLI 2009, *Aerosol Science and Technology*, 45, 46–64, doi:10.1080/02786826.2010.517813, 2011.
- Brioude, J., Arnold, D., Stohl, A., Cassiani, M., Morton, D., Seibert, P., Angevine, W., Evan, S., Dingwell, A., Fast, J. D., Easter, R. C., Pissio, I., Burkhardt, J., and Wotawa, G.: The Lagrangian particle dispersion model FLEXPART-WRF version 3.1, *Geosci. Model Dev.*, 6, 1889–1904, doi:10.5194/gmd-6-1889-2013, 2013.
- Browse, J., Carslaw, K. S., Mann, G. W., Birch, C. E., Arnold, S. R., and Leck, C.: The complex response of Arctic aerosol to sea-ice retreat, *Atmos. Chem. Phys.*, 14, 7543–7557, doi:10.5194/acp-14-7543-2014, 2014.
- Bzdek, B. R., Lawler, M. J., Horan, A. J., Pennington, M. R., DePalma, J. W., Zhao, J., Smith, J. N., and Johnston, M. V.: Molecular constraints on particle growth during new particle formation, *Geophys. Res. Lett.*, 41, 6045–6054, doi:10.1002/2014GL060160, 2014.
- Cai, Y., Montague, D., Mooiweer-Bryan, W., and Deshler, T.: Performance characteristics of the ultra-high sensitivity aerosol spectrometer for particles between 55 and 800 nm: Laboratory and field studies, *J. Aerosol Sci.*, 39, 759–769, 2008.
- Canagaratna, M. R., Jimenez, J. L., Kroll, J. H., Chen, Q., Kessler, S. H., Massoli, P., Hildebrandt Ruiz, L., Fortner, E., Williams, L. R., Wilson, K. R., Surratt, J. D., Donahue, N. M., Jayne, J. T., and Worsnop, D. R.: Elemental ratio measurements of organic compounds using aerosol mass spectrometry: characterization, improved calibration, and implications, *Atmos. Chem. Phys.*, 15, 253–272, doi:10.5194/acp-15-253-2015, 2015.
- Cavalieri, D. J., Parkinson, C., Gloersen, P., and Zwally, H.: Sea Ice Concentrations from Nimbus-7 SMMR and DMSP SSM/I-SSMIS Passive Microwave Data, Boulder, Colorado USA: NASA National Snow and Ice Data Center Distributed Active Archive Center, doi:10.5067/8GQ8LZQVL0VL, 1996.
- Ceburnis, D., O'Dowd, C. D., Jennings, G. S., Facchini, M. C., Emblico, L., Decesari, S., Fuzzi, S., and Sakalys, J.: Marine aerosol chemistry gradients: Elucidating primary and secondary processes and fluxes, *Geophys. Res. Lett.*, 35, L07804, doi:10.1029/2008GL033462, 2008.
- Chang, R. Y.-W., Slowik, J. G., Shantz, N. C., Vlasenko, A., Liggio, J., Sjostedt, S. J., Leaitch, W. R., and Abbatt, J. P. D.: The hygroscopicity parameter (κ) of ambient organic aerosol at a field site subject to biogenic and anthropogenic influences: relationship to degree of aerosol oxidation, *Atmos. Chem. Phys.*, 10, 5047–5064, doi:10.5194/acp-10-5047-2010, 2010.
- Chang, R. Y.-W., Leck, C., Graus, M., Müller, M., Paatero, J., Burkhardt, J. F., Stohl, A., Orr, L. H., Hayden, K., Li, S.-M., Hansel, A., Tjernström, M., Leaitch, W. R., and Abbatt, J. P. D.: Aerosol composition and sources in the central Arctic Ocean during ASCOS, *Atmos. Chem. Phys.*, 11, 10619–10636, doi:10.5194/acp-11-10619-2011, 2011a.
- Chang, R. Y.-W., Sjostedt, S. J., Pierce, J. R., Papakyriakou, T. N., Scarratt, M. G., Michaud, S., Levasseur, M., Leaitch, W. R., and Abbatt, J. P. D.: Relating atmospheric and oceanic DMS levels to particle nucleation events in the Canadian Arctic, *J. Geophys. Res.-Atmos.*, 116, D00S03, doi:10.1029/2011JD015926, 2011b.
- Chhabra, P. S., Ng, N. L., Canagaratna, M. R., Corrigan, A. L., Russell, L. M., Worsnop, D. R., Flagan, R. C., and Seinfeld, J. H.: Elemental composition and oxidation of chamber organic aerosol, *Atmos. Chem. Phys.*, 11, 8827–8845, doi:10.5194/acp-11-8827-2011, 2011.
- Claeys, M., Wang, W., Vermeylen, R., Kourtev, I., Chi, X., Farhat, Y., Surratt, J., Gomez-Gonzalez, Y., Sciare, J., and Maenhaut, W.: Chemical characterisation of marine aerosol at Amsterdam Island during the austral summer of 2006–2007, *J. Aerosol Sci.*, 41, 13–22, doi:10.1016/j.jaerosci.2009.08.003, 2010.
- Clarke, A. D., Owens, S. R., and Zhou, J.: An ultrafine sea-salt flux from breaking waves: Implications for cloud condensation nuclei in the remote marine atmosphere, *J. Geophys. Res.-Atmos.*, 111, D06202, doi:10.1029/2005JD006565, 2006.
- Croft, B., Martin, R. V., Leaitch, W. R., Tunved, P., Breider, T. J., D'Andrea, S. D., and Pierce, J. R.: Processes controlling the annual cycle of Arctic aerosol number and size distributions, *Atmos. Chem. Phys.*, 16, 3665–3682, doi:10.5194/acp-16-3665-2016, 2016.
- Curry, J. A.: Interactions among aerosols, clouds and climate of the Arctic Ocean, *Sci. Total Environ.*, 160, 777–791, 1995.

- Dall'Osto, M., Ceburnis, D., Monahan, C., Worsnop, D. R., Bialek, J., Kulmala, M., Kurtén, T., Ehn, M., Wenger, J., Sodeau, J., Healy, R., and O'Dowd, C.: Nitrogenated and aliphatic organic vapors as possible drivers for marine secondary organic aerosol growth, *J. Geophys. Res.-Atmos.*, 117, D12311, doi:10.1029/2012JD017522, 2012.
- DeCarlo, P. F., Kimmel, J. R., Trimborn, A., Northway, M. J., Jayne, J. T., Aiken, A. C., Gonin, M., Fuhrer, K., Horvath, T., Docherty, K. S., Worsnop, D. R., and Jimenez, J. L.: Field-deployable, high-resolution, time-of-flight aerosol mass spectrometer, *Anal. Chem.*, 78, 8281–8289, 2006.
- DeCarlo, P. F., Dunlea, E. J., Kimmel, J. R., Aiken, A. C., Sueper, D., Crounse, J., Wennberg, P. O., Emmons, L., Shinozuka, Y., Clarke, A., Zhou, J., Tomlinson, J., Collins, D. R., Knapp, D., Weinheimer, A. J., Montzka, D. D., Campos, T., and Jimenez, J. L.: Fast airborne aerosol size and chemistry measurements above Mexico City and Central Mexico during the MILAGRO campaign, *Atmos. Chem. Phys.*, 8, 4027–4048, doi:10.5194/acp-8-4027-2008, 2008.
- Decesari, S., Finessi, E., Rinaldi, M., Paglione, M., Fuzzi, S., Stephanou, E. G., Tziaras, T., Spyros, A., Ceburnis, D., O'Dowd, C., Dall'Osto, M., Harrison, R. M., Allan, J., Coe, H., and Facchini, M. C.: Primary and secondary marine organic aerosols over the North Atlantic Ocean during the MAP experiment, *J. Geophys. Res.-Atmos.*, 116, D22210, doi:10.1029/2011JD016204, 2011.
- Ehn, M., Thornton, J. A., Kleist, E., Sipila, M., Junninen, H., Pullinen, I., Springer, M., Rubach, F., Tillmann, R., Lee, B., Lopez-Hilfiker, F., Andres, S., Acir, I.-H., Rissanen, M., Jokinen, T., Schobesberger, S., Kangasluoma, J., Kontkanen, J., Nieminen, T., Kurten, T., Nielsen, L. B., Jorgensen, S., Kjaergaard, H. G., Canagaratna, M., Maso, M. D., Berndt, T., Petaja, T., Wahner, A., Kerminen, V.-M., Kulmala, M., Worsnop, D. R., Wildt, J., and Mentel, T. F.: A large source of low-volatility secondary organic aerosol, *Nature*, 506, 476–479, doi:10.1038/nature13032, 2014.
- Engvall, A.-C., Krejci, R., Ström, J., Treffeisen, R., Scheele, R., Hermansen, O., and Paatero, J.: Changes in aerosol properties during spring-summer period in the Arctic troposphere, *Atmos. Chem. Phys.*, 8, 445–462, doi:10.5194/acp-8-445-2008, 2008.
- Facchini, M. C., Decesari, S., Rinaldi, M., Carbone, C., Finessi, E., Mihaela, M., Fuzzi, S., Moretti, F., Tagliavini, E., Ceburnis, D., and O'Dowd, C.: Important Source of Marine Secondary Organic Aerosol from Biogenic Amines, *Environ. Sci. Technol.*, 42, 9116–9121, doi:10.1021/es8018385, 2008a.
- Facchini, M. C., Rinaldi, M., Decesari, S., Carbone, C., Finessi, E., Mircea, M., Fuzzi, S., Ceburnis, D., Flanagan, R., Nilsson, E. D., de Leeuw, G., Martino, M., Woeltjen, J., and O'Dowd, C. D.: Primary submicron marine aerosol dominated by insoluble organic colloids and aggregates, *Geophys. Res. Lett.*, 35, L17814, doi:10.1029/2008GL034210, 2008b.
- Friedl, M., Sulla-Menashe, D., Tan, B., Schneider, A., Ramankutty, N., Sibley, A., and Huang, X.: MODIS Collection 5 global land cover: Algorithm refinements and characterization of new datasets, Collection 5.1 IGBP Land Cover, Boston University, Boston, MA, USA, available at: <http://glcf.umd.edu/data/1c/> (last access: 31 May 2016), 2010.
- Frossard, A. A., Russell, L. M., Burrows, S. M., Elliott, S. M., Bates, T. S., and Quinn, P. K.: Sources and composition of submicron organic mass in marine aerosol particles, *J. Geophys. Res.-Atmos.*, 119, 12977–13003, 2014.
- Fu, P. Q., Kawamura, K., Chen, J., and Barrie, L. A.: Isoprene, Monoterpene, and Sesquiterpene Oxidation Products in the High Arctic Aerosols during Late Winter to Early Summer, *Environ. Sci. Technol.*, 43, 4022–4028, 2009.
- Fu, P. Q., Kawamura, K., Chen, J., Charrière, B., and Sempéré, R.: Organic molecular composition of marine aerosols over the Arctic Ocean in summer: contributions of primary emission and secondary aerosol formation, *Biogeosciences*, 10, 653–667, doi:10.5194/bg-10-653-2013, 2013.
- Fu, P. Q., Kawamura, K., Chen, J., Qin, M., Ren, L., Sun, Y., Wang, Z., Barrie, L., Tachibana, E., Ding, A., and Yamashita, Y.: Fluorescent water-soluble organic aerosols in the High Arctic atmosphere, *Scientific Reports*, 5, 9845, doi:10.1038/srep09845, 2015.
- Gantt, B. and Meskhidze, N.: The physical and chemical characteristics of marine primary organic aerosol: a review, *Atmos. Chem. Phys.*, 13, 3979–3996, doi:10.5194/acp-13-3979-2013, 2013.
- Gantt, B., Meskhidze, N., and Kamykowski, D.: A new physically-based quantification of marine isoprene and primary organic aerosol emissions, *Atmos. Chem. Phys.*, 9, 4915–4927, doi:10.5194/acp-9-4915-2009, 2009.
- Gao, R. S., Schwarz, J. P., Kelly, K. K., Fahey, D. W., Watts, L. A., Thompson, T. L., Spackman, J. R., Slowik, J. G., Cross, E. S., Han, J.-H., Davidovits, P., Onasch, T. B., and Worsnop, D. R.: A Novel Method for Estimating Light-Scattering Properties of Soot Aerosols Using a Modified Single-Particle Soot Photometer, *Aerosol Sci. Tech.*, 41, 125–135, doi:10.1080/02786820601118398, 2007.
- Garrett, T. J., Brattström, S., Sharma, S., Worthy, D. E. J., and Novelli, P.: The role of scavenging in the seasonal transport of black carbon and sulfate to the Arctic, *Geophys. Res. Lett.*, 38, L16805, doi:10.1029/2011GL048221, 2011.
- Giamarelou, M., Eleftheriadis, K., Nyeki, S., Tunved, P., Tørseth, K., and Biskos, G.: Indirect evidence of the composition of nucleation mode atmospheric particles in the high Arctic, *J. Geophys. Res.-Atmos.*, 121, 965–975, doi:10.1002/2015JD023646, 2016.
- Gosselin, M., Charette, J., Blais, M., Gourdau, M., Lizotte, M., Levasseur, M., Tremblay, J., and Gratton, Y.: Phytoplankton dynamics at receding ice edges in the Canadian High Arctic, 3rd Annual NETCARE Workshop, University of Toronto, Toronto, Ontario, Canada, 16–17 November 2015, available at: <http://www.netcare-project.ca/workshops/netcare-workshop-2015/> (last access: 18 February 2016), 2015.
- Grannas, A. M., Jones, A. E., Dibb, J., Ammann, M., Anastasio, C., Beine, H. J., Bergin, M., Bottenheim, J., Boxe, C. S., Carver, G., Chen, G., Crawford, J. H., Dominé, F., Frey, M. M., Guzmán, M. I., Heard, D. E., Helmig, D., Hoffmann, M. R., Honrath, R. E., Huey, L. G., Hutterli, M., Jacobi, H. W., Klán, P., Lefer, B., McConnell, J., Plane, J., Sander, R., Savarino, J., Shepson, P. B., Simpson, W. R., Sodeau, J. R., von Glasow, R., Weller, R., Wolff, E. W., and Zhu, T.: An overview of snow photochemistry: evidence, mechanisms and impacts, *Atmos. Chem. Phys.*, 7, 4329–4373, doi:10.5194/acp-7-4329-2007, 2007.
- Hansen, A. M. K., Kristensen, K., Nguyen, Q. T., Zare, A., Cozzi, F., Nøjgaard, J. K., Skov, H., Brandt, J., Christensen, J. H., Ström, J., Tunved, P., Krejci, R., and Glasius, M.: Organosulfates

- and organic acids in Arctic aerosols: speciation, annual variation and concentration levels, *Atmos. Chem. Phys.*, 14, 7807–7823, doi:10.5194/acp-14-7807-2014, 2014.
- Hayden, K. L., Sills, D. M. L., Brook, J. R., Li, S.-M., Makar, P. A., Markovic, M. Z., Liu, P., Anlauf, K. G., O'Brien, J. M., Li, Q., and McLaren, R.: Aircraft study of the impact of lake-breeze circulations on trace gases and particles during BAQS-Met 2007, *Atmos. Chem. Phys.*, 11, 10173–10192, doi:10.5194/acp-11-10173-2011, 2011.
- Healy, R. M., Evans, G. J., Murphy, M., Sierau, B., Arndt, J., McGillicuddy, E., O'Connor, I. P., Sodeau, J. R., and Wenger, J. C.: Single-particle speciation of alkylamines in ambient aerosol at five European sites, *Anal. Bioanal. Chem.*, 407, 5899–5909, 2015.
- Heintzenberg, J. and Leck, C.: The summer aerosol in the central Arctic 1991–2008: did it change or not?, *Atmos. Chem. Phys.*, 12, 3969–3983, doi:10.5194/acp-12-3969-2012, 2012.
- Heintzenberg, J., Leck, C., and Tunved, P.: Potential source regions and processes of aerosol in the summer Arctic, *Atmos. Chem. Phys.*, 15, 6487–6502, doi:10.5194/acp-15-6487-2015, 2015.
- Herber, A., Dethloff, K., Haas, C., Steinhage, D., Strapp, J. W., Bottenheim, J., McElroy, T., and Yamanouchi, T.: POLAR 5 – a new research aircraft for improved access to the Arctic, ISAR-1, Drastic Change under the Global Warming, Extended Abstract, 54–57, doi:10013/epic.34660, 2008.
- Hinz, K.-P., Greweling, M., Drews, F., and Spengler, B.: Data processing in on-line laser mass spectrometry of inorganic, organic, or biological airborne particles, *J. Am. Soc. Mass Spectr.*, 10, 648–660, doi:10.1016/S1044-0305(99)00028-8, 1999.
- Intrieri, J. M., Fairall, C. W., Shupe, M. D., Persson, P. O. G., Andreas, E. L., Guest, P. S., and Moritz, R. E.: An annual cycle of Arctic surface cloud forcing at SHEBA, *J. Geophys. Res.-Oceans*, 107, 8039, doi:10.1029/2000JC000439, 2002.
- Jeffries, M. and Richter-Menge, J.: The Arctic, in: State of the Climate in 2011, *B. Am. Meteorol. Soc.*, 93, 127–147, doi:10.1175/2012BAMSStateoftheClimate.1, 2012.
- Jimenez, J. L., Jayne, J. T., Shi, Q., Kolb, C. E., Worsnop, D. R., Yourshaw, I., Seinfeld, J. H., Flagan, R. C., Zhang, X., Smith, K. A., Morris, J. W., and Davidovits, P.: Ambient aerosol sampling using the Aerodyne Aerosol Mass Spectrometer, *J. Geophys. Res.-Atmos.*, 108, 8425, doi:10.1029/2001JD001213, 2003.
- Karl, M., Leck, C., Gross, A., and Pirjola, L.: A study of new particle formation in the marine boundary layer over the central Arctic Ocean using a flexible multicomponent aerosol dynamic model, *Tellus B*, 64, 17158, doi:10.3402/tellusb.v64i0.17158, 2012.
- Karl, M., Leck, C., Coz, E., and Heintzenberg, J.: Marine nanogels as a source of atmospheric nanoparticles in the high Arctic, *Geophys. Res. Lett.*, 40, 3738–3743, doi:10.1002/grl.50661, 2013.
- Kawamura, K., Ono, K., Tachibana, E., Charrière, B., and Sempéré, R.: Distributions of low molecular weight dicarboxylic acids, ketoacids and α -dicarbonyls in the marine aerosols collected over the Arctic Ocean during late summer, *Biogeosciences*, 9, 4725–4737, doi:10.5194/bg-9-4725-2012, 2012.
- Kay, J. E. and Gettelman, A.: Cloud influence on and response to seasonal Arctic sea ice loss, *J. Geophys. Res.-Atmos.*, 114, D18204, doi:10.1029/2009JD011773, 2009.
- Kopeck, B. G., Feng, X., Michel, F. A., and Posmentier, E. S.: Influence of sea ice on Arctic precipitation, *P. Natl. Acad. Sci.*, 113, 46–51, doi:10.1073/pnas.1504633113, 2016.
- Kulmala, M. and Kerminen, V.-M.: On the formation and growth of atmospheric nanoparticles, *Atmos. Res.*, 90, 132–150, doi:10.1016/j.atmosres.2008.01.005, 2008.
- Laborde, M., Schnaiter, M., Linke, C., Saathoff, H., Naumann, K.-H., Möhler, O., Berlenz, S., Wagner, U., Taylor, J. W., Liu, D., Flynn, M., Allan, J. D., Coe, H., Heimerl, K., Dahlkötter, F., Weinzierl, B., Wollny, A. G., Zannata, M., Cozic, J., Laj, P., Hitznerberger, R., Schwarz, J. P., and Gysel, M.: Single Particle Soot Photometer intercomparison at the AIDA chamber, *Atmos. Meas. Tech.*, 5, 3077–3097, doi:10.5194/amt-5-3077-2012, 2012.
- Law, K. S. and Stohl, A.: Arctic air pollution: Origins and impacts, *Science*, 315, 1537–1540, 2007.
- Lawler, M. J., Whitehead, J., O'Dowd, C., Monahan, C., McFiggans, G., and Smith, J. N.: Composition of 15–85 nm particles in marine air, *Atmos. Chem. Phys.*, 14, 11557–11569, doi:10.5194/acp-14-11557-2014, 2014.
- Leaith, W. R., Sharma, S., Huang, L., Toom-Saunty, D., Chivulescu, A., Macdonald, A. M., von Salzen, K., Pierce, J. R., Bertram, A. K., Schroder, J. C., Shantz, N. C., Chang, R. Y.-W., and Norman, A.-L.: Dimethyl sulfide control of the clean summertime Arctic aerosol and cloud, *Elementa*, 1, 00017, doi:10.12952/journal.elementa.000017, 2013.
- Leaith, W. R., Korolev, A., Aliabadi, A. A., Burkart, J., Willis, M., Abbatt, J. P. D., Bozem, H., Hoor, P., Köllner, F., Schneider, J., Herber, A., Konrad, C., and Brauner, R.: Effects of 20–100 nanometre particles on liquid clouds in the clean summertime Arctic, *Atmos. Chem. Phys. Discuss.*, doi:10.5194/acp-2015-999, in review, 2016.
- Lindsay, R., Zhang, J., Steele, M., and Stern, H.: Arctic sea-ice retreat in 2007 follows thinning trend, *J. Climate*, 22, 165–176, doi:10.1175/2008JCLI2521.1, 2009.
- Lubin, D. and Vogelmann, A. M.: A climatologically significant aerosol longwave indirect effect in the Arctic, *Nature*, 439, 453–456, 2006.
- Mahajan, A. S., Shaw, M., Oetjen, H., Hornsby, K. E., Carpenter, L. J., Kaleschke, L., Tian-Kunze, X., Lee, J. D., Moller, S. J., Edwards, P., Commane, R., Ingham, T., Heard, D. E., and Plane, J. M. C.: Evidence of reactive iodine chemistry in the Arctic boundary layer, *J. Geophys. Res.-Atmos.*, 115, D20303, doi:10.1029/2009JD013665, 2010.
- Mauritsen, T., Sedlar, J., Tjernström, M., Leck, C., Martin, M., Shupe, M., Sjogren, S., Sierau, B., Persson, P. O. G., Brooks, I. M., and Swietlicki, E.: An Arctic CCN-limited cloud-aerosol regime, *Atmos. Chem. Phys.*, 11, 165–173, doi:10.5194/acp-11-165-2011, 2011.
- Metzger, A., Verheggen, B., Dommen, J., Duplissy, J., Prevot, A. S. H., Weingartner, E., Riipinen, I., Kulmala, M., Spracklen, D. V., Carslaw, K. S., and Baltensperger, U.: Evidence for the role of organics in aerosol particle formation under atmospheric conditions, *P. Natl. Acad. Sci.*, 107, 6646–6651, doi:10.1073/pnas.0911330107, 2010.
- Middlebrook, A. M., Bahreini, R., Jimenez, J. L., and Canagaratna, M. R.: Evaluation of Composition-Dependent Collection Efficiencies for the Aerodyne Aerosol Mass Spec-

- trometer using Field Data, *Aerosol Sci. Tech.*, 46, 258–271, doi:10.1080/02786826.2011.620041, 2012.
- Modini, R. L., Frossard, A. A., Ahlm, L., Russell, L. M., Corrigan, C. E., Roberts, G. C., Hawkins, L. N., Schroder, J. C., Bertram, A. K., Zhao, R., Lee, A. K. Y., Abbatt, J. P. D., Lin, J., Nenes, A., Wang, Z., Wonaschütz, A., Sorooshian, A., Noone, K. J., Jonsen, H., Seinfeld, J. H., Toom-Sauntry, D., Macdonald, A. M., and Leaitch, W. R.: Primary marine aerosol-cloud interactions off the coast of California, *J. Geophys. Res.-Atmos.*, 120, 4282–4303, doi:10.1002/2014JD022963, 2015.
- Mungall, E. L., Croft, B., Lizotte, M., Thomas, J. L., Murphy, J. G., Levasseur, M., Martin, R. V., Wentzell, J. J. B., Liggio, J., and Abbatt, J. P. D.: Dimethyl sulfide in the summertime Arctic atmosphere: measurements and source sensitivity simulations, *Atmos. Chem. Phys.*, 16, 6665–6680, doi:10.5194/acp-16-6665-2016, 2016.
- Murphy, D. M.: The effects of molecular weight and thermal decomposition on the sensitivity of a thermal desorption aerosol mass spectrometer, *Aerosol Sci. Tech.*, 50, 118–125, doi:10.1080/02786826.2015.1136403, 2015.
- Murphy, D. M., Thomson, D. S., and Middlebrook, A. M.: Bromine, iodine, and chlorine in single aerosol particles at Cape Grim, *Geophys. Res. Lett.*, 24, 3197–3200, 1997.
- Narukawa, M., Kawamura, K., Li, S.-M., and Bottenheim, J. W.: Stable carbon isotopic ratios and ionic composition of the high-Arctic aerosols: An increase in $\delta^{13}\text{C}$ values from winter to spring, *J. Geophys. Res.-Atmos.*, 113, D02312, doi:10.1029/2007JD008755, 2008.
- NETCARE (Network on Climate and Aerosols): Addressing Key Uncertainties in Remote Canadian Environments, available at: <http://www.netcare-project.ca> (last access: 18 June 2016), 2015.
- Nguyen, Q. T., Glasius, M., Sørensen, L. L., Jensen, B., Skov, H., Birmili, W., Wiedensohler, A., Kristensson, A., Nøjgaard, J. K., and Massling, A.: Seasonal variation of atmospheric particle number concentrations, new particle formation and atmospheric oxidation capacity at the high Arctic site Villum Research Station, Station Nord, *Atmos. Chem. Phys. Discuss.*, doi:10.5194/acp-2016-205, in review, 2016.
- Nilsson, E. D., Rannik, U., Swietlicki, E., Leck, C., Aalto, P. P., Zhou, J., and Norman, M.: Turbulent aerosol fluxes over the Arctic Ocean 2. Wind-driven sources from the sea, *J. Geophys. Res.-Atmos.*, 106, 32139–32154, 2001.
- O'Dowd, C. and de Leeuw, G.: Marine aerosol production: a review of current knowledge, *Philos. T. Roy. Soc.*, 365, 1753–1774, doi:10.1098/rsta.2007.2043, 2007.
- O'Dowd, C., Ceburnis, D., Ovadnevaite, J., Bialek, J., Stengel, D., Zacharias, M., Nitschke, U., Connan, S., Rinaldi, M., Fuzzi, S., Decesair, S., Facchini, M., Maullo, S., Santolieri, R., Dell'Anno, A., Corinaldesi, C., Tangherlini, M., and Danovaro, R.: Connecting marine productivity to sea-spray via nanoscale biological processes: Phytoplankton Dance or Death Disco?, *Scientific Reports*, 5, 14883, doi:10.1038/srep14883, 2015.
- Orellana, M. V., Matrai, P. A., Leck, C., Rauschenberg, C. D., Lee, A. M., and Coz, E.: Marine microgels as a source of cloud condensation nuclei in the high Arctic, *P. Natl. Acad. Sci. USA*, 108, 13612–13617, 2011.
- Ovadnevaite, J., Ceburnis, D., Martucci, G., Bialek, J., Monahan, C., Rinaldi, M., Facchini, M. C., Berresheim, H., Worsnop, D. R., and O'Dowd, C.: Primary marine organic aerosol: A dichotomy of low hygroscopicity and high CCN activity, *Geophys. Res. Lett.*, 38, L21806, doi:10.1029/2011GL048869, 2011.
- Ovadnevaite, J., Ceburnis, D., Canagaratna, M., Berresheim, H., Bialek, J., Martucci, G., Worsnop, D. R., and O'Dowd, C.: On the effect of wind speed on submicron sea salt mass concentrations and source fluxes, *J. Geophys. Res.-Atmos.*, 117, D16201, doi:10.1029/2011JD017379, 2012.
- Ovadnevaite, J., Manders, A., de Leeuw, G., Ceburnis, D., Monahan, C., Partanen, A.-I., Korhonen, H., and O'Dowd, C. D.: A sea spray aerosol flux parameterization encapsulating wave state, *Atmos. Chem. Phys.*, 14, 1837–1852, doi:10.5194/acp-14-1837-2014, 2014.
- Petters, M. D. and Kreidenweis, S. M.: A single parameter representation of hygroscopic growth and cloud condensation nucleus activity, *Atmos. Chem. Phys.*, 7, 1961–1971, doi:10.5194/acp-7-1961-2007, 2007.
- Phinney, L., Leaitch, W. R., Lohmann, U., Boudries, H., Worsnop, D. R., Jayne, J. T., Toom-Sauntry, D., Wadleigh, M., Sharma, S., and Shantz, N.: Characterization of the aerosol over the subarctic north east Pacific Ocean, *Deep Sea Res. Pt. II*, 53, 2410–2433, doi:10.1016/j.dsr2.2006.05.044, 2006.
- Quinn, P. K., Shaw, G., Andrews, E., Dutton, E. G., Ruoho-Airola, T., and Gong, S. L.: Arctic haze: current trends and knowledge gaps, *Tellus B*, 59, 99–114, 2007.
- Quinn, P. K., Bates, T., Schulz, K., Coffman, D., Frossard, A., Russell, L., Keene, W., and Kieber, D.: Contribution of sea surface carbon pool to organic matter enrichment in sea spray aerosol, *Nat. Geosci.*, 7, 228–232, doi:10.1038/NNGEO2092, 2015a.
- Quinn, P. K., Collins, D. B., Grassian, V. H., Prather, K. A., and Bates, T. S.: Chemistry and Related Properties of Freshly Emitted Sea Spray Aerosol, *Chem. Rev.*, 115, 4383–4399, doi:10.1021/cr500713g, 2015b.
- Rehbein, P. J. G., Jeong, C.-H., McGuire, M. L., Yao, X., Corbin, J. C., and Evans, G. J.: Cloud and Fog Processing Enhanced Gas-to-Particle Partitioning of Trimethylamine, *Environ. Sci. Technol.*, 45, 4346–4352, doi:10.1021/es1042113, 2011.
- Rinaldi, M., Decesari, S., Finessi, E., Giulianelli, L., Carbone, C., Fuzzi, S., O'Dowd, C. D., Ceburnis, D., and Facchini, M. C.: Primary and Secondary Organic Marine Aerosol and Oceanic Biological Activity: Recent Results and New Perspectives for Future Studies, *Advances in Meteorology*, 2010, 310682, doi:10.1155/2010/310682, 2010.
- Robinson, N. H., Hamilton, J. F., Allan, J. D., Langford, B., Oram, D. E., Chen, Q., Docherty, K., Farmer, D. K., Jimenez, J. L., Ward, M. W., Hewitt, C. N., Barley, M. H., Jenkin, M. E., Rickard, A. R., Martin, S. T., McFiggans, G., and Coe, H.: Evidence for a significant proportion of Secondary Organic Aerosol from isoprene above a maritime tropical forest, *Atmos. Chem. Phys.*, 11, 1039–1050, doi:10.5194/acp-11-1039-2011, 2011.
- Roth, A., Schneider, J., Klimach, T., Mertes, S., van Pinxteren, D., Herrmann, H., and Borrmann, S.: Aerosol properties, source identification, and cloud processing in orographic clouds measured by single particle mass spectrometry on a central European mountain site during HCCT-2010, *Atmos. Chem. Phys.*, 16, 505–524, doi:10.5194/acp-16-505-2016, 2016.
- Russell, L. M., Hawkins, L. N., Frossard, A. A., Quinn, P. K., and Bates, T. S.: Carbohydrate-like composition of submicron atmospheric particles and their production from

- ocean bubble bursting, *P. Natl. Acad. Sci.*, 107, 6652–6657, doi:10.1073/pnas.0908905107, 2010.
- Schollert, M., Burchard, S., Faubert, P., Michelsen, A., and Rinnan, R.: Biogenic volatile organic compound emissions in four vegetation types in high arctic Greenland, *Polar Biol.*, 37, 237–249, 2014.
- Schwarz, J. P., Gao, R. S., Fahey, D. W., Thomson, D. S., Watts, L. A., Wilson, J. C., Reeves, J. M., Darbeheshti, M., Baumgardner, D. G., Kok, G. L., Chung, S. H., Schulz, M., Hendricks, J., Lauer, A., Kärcher, B., Slowik, J. G., Rosenlof, K. H., Thompson, T. L., Langford, A. O., Loewenstein, M., and Aikin, K. C.: Single-particle measurements of midlatitude black carbon and light-scattering aerosols from the boundary layer to the lower stratosphere, *J. Geophys. Res.-Atmos.*, 111, D16207, doi:10.1029/2006JD007076, 2006.
- Seuper, D.: ToF-AMS analysis software, available at: <http://cires.colorado.edu/jimenez-group/ToFAMSResources/ToFSoftware/index.html> (last access: 20 November 2015), 2010.
- Sharma, S., Ishizawa, M., Chan, D., Lavoue, D., Andrews, E., Eleftheriadis, K., and Maksyutov, S.: 16-year simulation of Arctic black carbon: Transport, source contribution, and sensitivity analysis on deposition, *J. Geophys. Res.-Atmos.*, 118, 943–964, 2013.
- Shaw, P. M., Russell, L. M., Jefferson, A., and Quinn, P. K.: Arctic organic aerosol measurements show particles from mixed combustion in spring haze and from frost flowers in winter, *Geophys. Res. Lett.*, 37, L10803, doi:10.1029/2010GL042831, 2010.
- Shaw, S., Gantt, B., and Meskhidze, N.: Production and Emissions of Marine Isoprene and Monoterpenes: A Review, *Advances in Meteorology*, 2010, 408696, doi:10.1155/2010/408696, 2010.
- Skamarock, W. C., Klemp, J. B., Dudhia, J., Gill, D. O., Barker, D. M., Wang, W., and Powers, J. G.: A Description of the Advanced Research WRF Version 2, NCAR, Boulder, CO, USA, 88, 7–25, 2001.
- Stohl, A.: Characteristics of atmospheric transport into the Arctic troposphere, *J. Geophys. Res.-Atmos.*, 111, D11306, doi:10.1029/2005JD006888, 2006.
- Stohl, A., Forster, C., Frank, A., Seibert, P., and Wotawa, G.: Technical note: The Lagrangian particle dispersion model FLEXPART version 6.2, *Atmos. Chem. Phys.*, 5, 2461–2474, doi:10.5194/acp-5-2461-2005, 2005.
- Tjernström, M., Shupe, M. D., Brooks, I. M., Persson, P. O. G., Prytherch, J., Salisbury, D. J., Sedlar, J., Achtert, P., Brooks, B. J., Johnston, P. E., Sotiropoulou, G., and Wolfe, D.: Warm-air advection, air mass transformation and fog causes rapid ice melt, *Geophys. Res. Lett.*, 42, 5594–5602, doi:10.1002/2015GL064373, 2015.
- Tröstl, J., Chuang, W. K., Gordon, H., Heinritzi, M., Yan, C., Molteni, U., Ahlm, L., Frege, C., Bianchi, F., Wagner, R., Simon, M., Lehtipalo, K., Williamson, C., Craven, J. S., Duplissy, J., Adamov, A., Almeida, J., Bernhammer, A.-K., Breitenlechner, M., Brilke, S., Dias, A., Ehrhart, S., Flagan, R. C., Franchin, A., Fuchs, C., Guida, R., Gysel, M., Hansel, A., Hoyle, C. R., Jokinen, T., Junninen, H., Kangasluoma, J., Keskinen, H., Kim, J., Krapf, M., Kürten, A., Laaksonen, A., Lawler, M., Leiminger, M., Mathot, S., Möhler, O., Nieminen, T., Onnela, A., Petäjä, T., Piel, F. M., Miettinen, P., Rissanen, M. P., Rondo, L., Sarnela, N., Schobesberger, S., Sengupta, K., Sipilä, M., Smith, J. N., Steiner, G., Tomé, A., Virtanen, A., Wagner, A. C., Weingartner, E., Wimmer, D., Winkler, P. M., Ye, P., Carslaw, K. S., Curtius, J., Dommen, J., Kirkby, J., Kulmala, M., Riipinen, I., Worsnop, D. R., Donahue, N. M., and Baltensperger, U.: The role of low-volatility organic compounds in initial particle growth in the atmosphere, *Nature*, 533, 527–531, doi:10.1038/nature18271, 2016.
- Tunved, P., Ström, J., and Krejci, R.: Arctic aerosol life cycle: linking aerosol size distributions observed between 2000 and 2010 with air mass transport and precipitation at Zeppelin station, Ny-Ålesund, Svalbard, *Atmos. Chem. Phys.*, 13, 3643–3660, doi:10.5194/acp-13-3643-2013, 2013.
- Vaattovaara, P., Huttunen, P. E., Yoon, Y. J., Joutsensaari, J., Lehtinen, K. E. J., O'Dowd, C. D., and Laaksonen, A.: The composition of nucleation and Aitken modes particles during coastal nucleation events: evidence for marine secondary organic contribution, *Atmos. Chem. Phys.*, 6, 4601–4616, doi:10.5194/acp-6-4601-2006, 2006.
- Wentworth, G. R., Murphy, J. G., Croft, B., Martin, R. V., Pierce, J. R., Côté, J.-S., Courchesne, I., Tremblay, J.-É., Gagnon, J., Thomas, J. L., Sharma, S., Toom-Sauntry, D., Chivulescu, A., Levasseur, M., and Abbatt, J. P. D.: Ammonia in the summertime Arctic marine boundary layer: sources, sinks, and implications, *Atmos. Chem. Phys.*, 16, 1937–1953, doi:10.5194/acp-16-1937-2016, 2016.
- Yakobi-Hancock, J. D., Ladino, L. A., Bertram, A. K., Huffman, J. A., Jones, K., Leaitch, W. R., Mason, R. H., Schiller, C. L., Toom-Sauntry, D., Wong, J. P. S., and Abbatt, J. P. D.: CCN activity of size-selected aerosol at a Pacific coastal location, *Atmos. Chem. Phys.*, 14, 12307–12317, doi:10.5194/acp-14-12307-2014, 2014.
- Zorn, S. R., Drewnick, F., Schott, M., Hoffmann, T., and Borrmann, S.: Characterization of the South Atlantic marine boundary layer aerosol using an aerodyne aerosol mass spectrometer, *Atmos. Chem. Phys.*, 8, 4711–4728, doi:10.5194/acp-8-4711-2008, 2008.